

CHAPTER 2

Acids, Bases and Salts

THIS CHAPTER INCLUDES

For thousands of years, people have known that vinegar, lemon juice and many other foods taste sour. However, it was not until a few hundred years ago that it was discovered why these things taste sour – because these all contain acids. The term acid comes from the Latin term *acidus*, which means “sour”. In the same manner, bases are recognised due to their bitter taste. But taste of anything does not help to identify the nature and all the properties of any substance. Hence, a number of theories were proposed to describe acids and bases. In this chapter we will study the fundamentals of acid/base chemistry and the properties of the salts produced during acid-base reaction.

ACIDS

Acids generally are those chemical substances which have a sour taste. Acids change the colour of blue litmus solution to red. If we cut a lemon with a knife and taste it, lemon is sour in taste like that some common fruits such as raw mango, raw grapes, orange and tamarind etc. An acid is a substance which contains hydrogen ions and produces H^+ ions in its aqueous solutions. (Arrhenius concept)

Classification of Acids

Acids

Depending on strength

- ⇨ Strong acids
- ⇨ Weak acids

Depending on sources

- ⇨ Organic acids
- ⇨ Inorganic acids

Depending on concentration

- ⇨ Concentrated acids
- ⇨ Dilute acids

(1) **Depending on Strength** : Strength of an acid means, to what extent dissociation of an acid takes place in the solution. This is known as “**degree of dissociation**” of an acid. It is represented by alpha (α).

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