

# Reversible Reactions & Chemical Equilibrium

- **Reversible reaction** = chemical reaction that can occur in either direction.
- Forward reaction:  $A \rightarrow B$
- Reverse reaction:  $A \leftarrow B$
- Usually combined:  $A \rightleftharpoons B$
- **Chemical equilibrium** = point at which the rate of forward reaction **equals** the rate of reverse reaction.

# Concept Review Section Reversible Reactions And Equilibrium

**Conrad L. Stanitski**



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