Concept Review

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10 11 12 13

- ion octet anion cation
- 1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n)

Complete each statement below by writing the correct term or phrase.

- 4. An anion is an ion with a regarive charge.
- 5. Valence electrons occupy the outermost energy level of an atom.
- 7. A cation is an ion with a possitive charge.
- 8. Many stable ions have an electron configuration of a octet

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

else is missing some

Concept Review Section Simple Ions Answers

Ralph Thomas Myers

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