- 1. Which conditions must be met for a reaction to take place?
 - I. Reactants collide with sufficient energy.
 - II. Reactants collide with correct orientation.
 - III. Reactants must be in the same state.

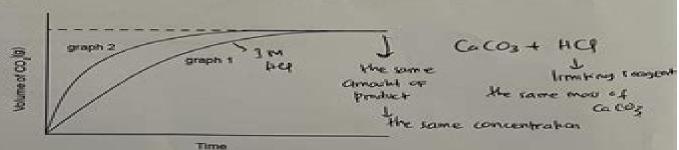
B. I and III only C. II and III only



Curves I and II represent samples of the same gas at a constant pressure but at different temperatures. The areas under curves I and II are equal. What does curve II represent?



- A. Curve II is at the lower temperature and there are less molecules in the sample.
- B. Curve II is at the lower temperature and there are the same number of molecules in the samples.
- C. Curve II is at the higher temperature and there are more molecules in the sample.
- Curve II is at the higher temperature and there are the same number of molecules in the samples.
- 3. Graph I shows a plot of volume of CO2 (g) against time for the reaction of CaCO3 (s) with 1.00 mol dm⁻³ HCI (aq). The acid is the limiting reagent and entirely covers the lumps of CaCO3 (s). Which set of conditions is most likely to give the data plotted in graph 2 when the same mass of CaCO3 (s) is reacted with the same volume of HCI (aq) at the same temperature?



Conceptual Multiple Chocie Refraction Questions

Kathleen P. Talaro, Arthur Talaro

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Instructor's Guide to Accompany Physics, Concepts and Applications, 2nd Edition Jerry D. Wilson, 1981 College Physics Essentials, Eighth Edition Jerry D. Wilson, Anthony J. Buffa, Bo Lou, 2019-12-06 This new edition of College Physics Essentials provides a streamlined update of a major textbook for algebra based physics This is the second volume covers electricity atomic nuclear and quantum physics The first volume is available separately and covers topics such as mechanics heat and thermodynamics The authors provide emphasis on worked examples together with expanded problem sets that build from conceptual understanding to numerical solutions and real world applications to increase reader engagement Including over 900 images throughout the two volumes this textbook is highly recommended for students seeking a basic understanding of key physics concepts and how to apply them to real problems **Photonic Engineering** Brian W. Bowe, 2005 Proceedings of SPIE present the original research papers presented at SPIE conferences and other high quality conferences in the broad ranging fields of optics and photonics These books provide prompt access to the latest innovations in research and technology in their respective fields Proceedings of SPIE are among the most cited references in patent Harcourt Science: Physical science, [grade] 5, Units E and F, teacher's ed ,2000 literature Instructor's Manual to Accompany Conceptual Physics Paul G. Hewitt, 1977 Physics Eugene Hecht, 2000-02 In Hecht s groundbreaking book you ll find real life applications and unparalleled art and photography program a presentation that

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