

**CHEM 101 Quiz #6 Chapter 6 ANSWERS****1. (4 points)**

The specific heat (s) of iron (Fe) is  $0.40 \text{ J/(g K)}$ . Calculate the heat effect  $q$  when 20 g of iron is heated from  $100^\circ\text{C}$  to  $200^\circ\text{C}$ . Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \Delta K = \Delta^\circ\text{C} \quad q = 20 \times 0.40 \times (200 - 100) = +800 \text{ J}$$

You must include the units and sign in the answer!

**2. (4 points)**

The Molar Mass of Mg =  $24.0 \text{ g/mol}$ .

Calculate the enthalpy change when 2.40 g Mg(s) reacts with excess oxygen. Include the sign!

$$2.40 \text{ g Mg} = 2.40/24.0 = 0.1 \text{ mol} \rightarrow \Delta H = 0.1(\text{mol}) \times (-600)(\text{kJ/mol}) = -60 \text{ kJ}$$

Again, your answer must include both the sign and the units.

**3. (2 points)** Which one of the following four reactions is a *formation reaction*? Circle the correct answer.



*Look up the definition of a formation reaction in Chang or in your lecture notes: the compound is formed from its elements, and the elements must be as they naturally occur so e.g.  $\text{Cl}_2$  not  $\text{Cl}$ , and  $\text{H}_2$  not  $\text{H}$ !*

**CHEM 101 Section L52 Quiz #6****1. (4 points)**

The specific heat (s) of copper (Cu) is  $0.38 \text{ J/(g K)}$ . Calculate the heat effect  $q$  when 10 g of copper is heated from  $100^\circ\text{C}$  to  $300^\circ\text{C}$ . Include the sign!

$$q = m \cdot s \cdot \Delta T \text{ for } \Delta T \Delta K = \Delta^\circ\text{C} \quad q = 10 \times 0.38 \times (300 - 100) = +760 \text{ J}$$

You must include the units and sign in the answer!

**2. (4 points)**

The Molar Mass of Ca =  $40.0 \text{ g/mol}$ .

Calculate the enthalpy change  $\Delta H$  when 8.0 g Ca(s) reacts with excess oxygen. Include the sign!!

$$8.0 \text{ g Ca} = 8.0/40 = 0.2 \text{ mol} \rightarrow \Delta H = 0.2(\text{mol}) \times (-650)(\text{kJ/mol}) = -130 \text{ kJ}$$

Again, your answer must include both the sign and the units.

**3. (2 points)** Which one of the following reactions is a *formation reaction*? Circle the correct answer.



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In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Core Teaching Resources Chemistry Answers Chapter 6 Quiz**," a mesmerizing literary creation penned by a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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### **Table of Contents Core Teaching Resources Chemistry Answers Chapter 6 Quiz**

1. Understanding the eBook Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - The Rise of Digital Reading Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Advantages of eBooks Over Traditional Books
2. Identifying Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - User-Friendly Interface
4. Exploring eBook Recommendations from Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Personalized Recommendations

- Core Teaching Resources Chemistry Answers Chapter 6 Quiz User Reviews and Ratings
- Core Teaching Resources Chemistry Answers Chapter 6 Quiz and Bestseller Lists
- 5. Accessing Core Teaching Resources Chemistry Answers Chapter 6 Quiz Free and Paid eBooks
  - Core Teaching Resources Chemistry Answers Chapter 6 Quiz Public Domain eBooks
  - Core Teaching Resources Chemistry Answers Chapter 6 Quiz eBook Subscription Services
  - Core Teaching Resources Chemistry Answers Chapter 6 Quiz Budget-Friendly Options
- 6. Navigating Core Teaching Resources Chemistry Answers Chapter 6 Quiz eBook Formats
  - ePub, PDF, MOBI, and More
  - Core Teaching Resources Chemistry Answers Chapter 6 Quiz Compatibility with Devices
  - Core Teaching Resources Chemistry Answers Chapter 6 Quiz Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Highlighting and Note-Taking Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Interactive Elements Core Teaching Resources Chemistry Answers Chapter 6 Quiz
- 8. Staying Engaged with Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Core Teaching Resources Chemistry Answers Chapter 6 Quiz
- 9. Balancing eBooks and Physical Books Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Core Teaching Resources Chemistry Answers Chapter 6 Quiz
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Setting Reading Goals Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Core Teaching Resources Chemistry Answers Chapter 6 Quiz
  - Fact-Checking eBook Content of Core Teaching Resources Chemistry Answers Chapter 6 Quiz

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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