Dilutions Worksheet

- If I have 340 mL of a 1.5 M NaBr solution, what will the concentration be if I add 560 mL more water to it?
- 2) If I dilute 250 mL of 0.50 M lithium acetate solution to a volume of 750 mL, what will the concentration of this solution be?
- 3) If I leave 750 mL of 0.75 M sodium chloride solution uncovered on a windowsill and 150 mL of the solvent evaporates, what will the new concentration of the sodium chloride solution be?
- 4) To what volume would I need to add water to the evaporated solution in problem 3 to get a solution with a concentration of 0.50 M?
- 5) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?
- 6) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?
- 7) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

Dilution Notes And Practice Answers

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