# Experiment 7 The Solution is Dilution REPORT

Fill in this table using the instructions in the lab manual and from video:

Test Tube#	mL Stock Solution	mL DI water	Total Volume of solution (mL)	Molarity of solution, M
1	10.0	0.0		0 - 20
3	8.0	2.0		0-14
3	6.0	4.0		0.13
4	4.0	6.0		0.08
5	2.0	6.0		0.04

Complete the "Student analysis of colorimeter data and unknown" page of this experiment.

What is the concentration of the unknown solution of nickel(II) chloride?

Questions needing solutions:

a = 2 Cs

Use the lab manual to find the wavelength of light absorbed and the color wheel (previous experiment) to fill in the table below.

Solution	Color of light absorbed	Color of solution
Co(NO <sub>3</sub> ) <sub>2</sub> (aq)		
NiCl <sub>2</sub> (aq)		

What is the relationship between the color of light absorbed by the solution and the color of the solution? (Explain below)

- Suppose 25.00 g of C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>(sucrose) is added to a 50.00 mL volumetric flask. DI water is added to the flask until the total volume of the solution is 50.00 mL. What is the molarity of the resulting solution?

# **Dilution Of Solution Lab Report**

**B. Pam Ismail, S. Suzanne Nielsen** 

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In a digital era where connections and knowledge reign supreme, the enchanting power of language has be much more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Dilution Of Solution Lab Report**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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