INTERNATIONAL INDIAN SCHOOL, RIYADH

SUMMATIVE II WORKSHEET

CLASS X

SUBJECT: CHEMISTRY

CHAPTER: CARBON AND ITS COMPOUND

- Why does carbon forms compounds mainly by covalent bonding?
- Name the three allotropes of carbon.
- 3. Which of the following compounds can have a triple bond?

 C_2H_4 C_3H_4 C_3H_6

4. Which of the following compounds can have a double bond?

C₄H₁₀ C₅H₈ C₅H₁₀

- Give the names of one saturated cyclic hydrocarbon and one unsaturated cyclic hydrocarbon.
- 6. Write the molecular formula and structure of benzene.
- 7. Write the electron dot structure for the following compounds:

(1) H₂ (2) O₂ (3) N₂ (4) CO₂ (5) S₈ (6) H₂O (7) NH₁ (8) CCI₄

(9)CH₄ (10) C_5H_{12} (11) C_4H_8 (12) C_5H_8 (13) C_7H_{12}

(14)C₅H₁₀ (15)Cyclopropane (16) Cyclobutane

(17)Cyclopentane (18)Cyclohexane (19)Benzene

- Define the following with examples:
 - (a) Hydrocarbon.
 - (b)Saturated hydrocarbon.
 - (c)Unsaturated hydrocarbon.
 - (d)Homologous series.

(e)Functional group.

- Write the IUPAC Name of the following:
 - (1) CH₃ CH₂ CH₂ CI
 - (2) C₃H₇Br
 - (3) C₂H₅I

Carbon Chemistry Chapter Test A

RP Manchanda

Carbon Chemistry Chapter Test A:

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,1902 **Foundry** ,1924 <u>Metallurgy of Cast Iron</u> Thomas Dyson West,1897

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