

Conversion of Ore to Metal Oxide

Calcination	Roasting
It is done for Carbonate Ores	It is done for Sulphide Ores
Heating of carbonate ores in absence of oxygen	Heating of Sulphide ores in presence of oxygen
CO ₂ gas is released and metal oxide is obtained	SO ₂ gas is released and metal oxide is obtained
$\text{ZnCO}_3 (\text{s}) \xrightarrow{\text{Heat}} \text{ZnO} (\text{s}) + \text{CO}_2 (\text{g})$	$\text{ZnS} (\text{s}) + 3\text{O}_2 \xrightarrow{\text{Heat}} 2\text{ZnO} (\text{s}) + 2\text{SO}_2 (\text{g})$

Carbonate Ores Experiment

Michael Rajnik

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

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