Equilibrium Quiz Choose the answer that nest fits each question

100	Once a chemical equilibrium is established the and of reaction are equal to each other.			
	27 045 EPS	final concentration	ns c reactor	t and product
		nd reverse rates processes	d. decony	position and
2.	Once chemical a longer a(n) a. Thermal	equilibrium of a re change in c b. small c. in	action has been re oncentrations of c significant	hached there is no hemical species. d. observable
36	which of the foll a. Ges	lawing do not affe b. Liquid	ct equilibrium? Ma c. Aqueous	ork all that apply d. Solid
200 			quilibrium shift? M n c. Temperature	
5.		ii. neactions iii. product		pared to 1
6.	Using the reacti- how much C will	on: A + B <> 2 be present once	C. If you have 4M the reaction reach	A and 3M 8 then es equilibrium? K =
	a 2.06 M	b. 7.36 M	c. 0.03 M	d. 4.18 M
7.	Using the reaction: $A+2B \leftarrow \sim 2C$. If you have 2M A and 2M B then how much 0 will be present once the reaction reaches equilibrium? $K\sim 1.0 \times 10^6$			
	a 1.004 M	b. 0.002814	c. 0.035 M	d. 0.00023 M
	Using the reaction: $3A+2B \ll 3C$. Calculate the equilibrium constant if the concentrations at equilibrium are $A=2M$, $B=3M$, and $C=1.5M$			
8.	constant if the c	oncentrations at	equilibrium are A	- 2M. 8 = 3M. and

Use the graph below to answer the questions

Chapter 1chemical Equilibrium Quiz

James E. Banks

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