

CHAPTER 14 STUDY GUIDE

Mixtures and Solutions

Section 14.1 Heterogeneous and Homogeneous Mixtures

In your textbook, read about suspensions and colloids.

For each statement below, write *true* or *false*.

- _____ 1. A solution is a mixture containing particles that settle out of the mixture if left undisturbed.
- _____ 2. The most abundant substance in a colloid is the dispersion medium.
- _____ 3. A colloid can be separated by filtration.
- _____ 4. A solid emulsion consists of a liquid dispersed in a solid.
- _____ 5. Whipped cream is an example of a foam.
- _____ 6. In an aerosol, the dispersing medium is a liquid.
- _____ 7. Brownian motion results from the collisions of particles of the dispersion medium with the dispersed particles.
- _____ 8. Dispersed particles in a colloid do not tend to settle out because they have polar or charged atomic groups on their surfaces.
- _____ 9. Stirring an electrolyte into a colloid stabilizes the colloid.
- _____ 10. Colloids demonstrate the Tyndall effect.

The table below lists the characteristics of particles in colloids, solutions, and suspensions. Place a check in the column of each mixture whose particles have a particular characteristic.

Characteristics of Particles	Colloid	Solution	Suspension
11. Less than 1 nm in diameter			
12. Between 1 nm and 1000 nm in diameter			
13. More than 1000 nm in diameter			
14. Settle out if undisturbed			
15. Pass through standard filter paper			
16. Lower vapor pressure			
17. Scatter light			

Chapter 1 study Guide For Content Mastery

Stoichiometry

Glencoe/McGraw-Hill



Chapter 1study Guide For Content Mastery Stoichiometry:

Resources in Education ,1980 Research in Education ,1969 *Study Guide for Chemical Principles [by] Steven S. Zumdahl* Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Glencoe Chemistry* Glencoe/McGraw-Hill,2002 Stoichiometry Review (Inorganic Chemistry Fast Facts) E Staff, Learn and review on the go Use Quick Review Chemistry Study Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Learn how to solve various reaction problems in Chemistry Perfect study notes for all high school health sciences premed medical and nursing students **You Can Do Chemistry** Sunil Tanna,2018-11-19 A comprehensive guide to performing mole and stoichiometric calculations with numerous examples as well as questions and answers Covers calculations relating to solids solutions gases and electrolysis plus as limiting and excess reactants chemical yields atom economy and much more Fully up to date with the last international standards including the revised definition of mole which was agreed on November 16th 2018

This book delves into Chapter 1study Guide For Content Mastery Stoichiometry. Chapter 1study Guide For Content Mastery Stoichiometry is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Chapter 1study Guide For Content Mastery Stoichiometry, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 1: Introduction to Chapter 1study Guide For Content Mastery Stoichiometry
 - Chapter 2: Essential Elements of Chapter 1study Guide For Content Mastery Stoichiometry
 - Chapter 3: Chapter 1study Guide For Content Mastery Stoichiometry in Everyday Life
 - Chapter 4: Chapter 1study Guide For Content Mastery Stoichiometry in Specific Contexts
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