

Chapter 2 Chemistry Assessment Answers

2.1 Assessment Answers

1. Intensive and extensive properties
2. Every sample of a given substance has the same chemical composition.
3. Solid, liquid, gas
4. Physical changes are either reversible or irreversible. Reversible changes can be "undone", or reversed. Irreversible changes cannot be undone.
5. Color; sodium chloride is the only white solid listed.
6. Liquids and gases have indefinite shape. The shape of a solid is definite; the shape of a liquid is indefinite.
7. The freezing of mercury is reversible because solid mercury can be melted.
8. Samples of gold and copper can have the same mass and volume (extensive properties). They cannot have the same set of intensive properties because they have different chemical compositions.

2.2 Assessment Answers

11. As heterogeneous or homogeneous
12. Differences in physical properties.
13. A phase is any part of a sample with uniform composition. There is one phase in a homogeneous mixture and two or more phases in a heterogeneous mixture.
14.
 - a. homogeneous
 - b. heterogeneous
 - c. homogeneous
 - d. heterogeneous
15. Both have a uniform composition throughout. A substance has a definite composition; a solution has a variable composition.
16. Filtrations separates solids from a liquid in a heterogeneous mixture. Distillation can separate a liquid from a substance dissolved in the liquid.
17. Add water to dissolve the salt. Pour the resulting mixture onto a piece of closely woven cloth. The sand will remain on the cloth, and the salt solution will pass through. Use evaporation to remove the water from the salt solution, leaving solid salt behind.

Chapter 2 Assessment Chemistry Answers

Alton Biggs



Chapter 2 Assessment Chemistry Answers:

Study Guide and Solutions Manual to accompany Basic Concepts of Chemistry, 9th Edition Leo J.

Malone,2011-12-07 This is the Study Guide and Solutions Manual to accompany Malone s Basic Concepts of Chemistry

FASTtrack: Chemistry of Drugs David Barlow,David Mountford,2014-07-02 A revision guide on pharmaceutical and medicinal chemistry The book covers all aspects of the chemistry of drugs and includes key points tips and self assessment questions to aid in learning *Prentice Hall Chemistry* ,2000 *Astronomy* Alton Biggs,2002 *Agrochemical Environmental Fate State of the Art* Marguerite L. Leng,Elizabeth M.K. Leovey,Paul L. Zubkoff,1995-04-28 Accurate risk assessment is critical to pesticide regulation This authoritative reference provides an exhaustive evaluation of current agrochemical environmental fate studies a critical review of current EPA pesticide assessment guidelines and a wide variety of environmental simulation models Divided into four sections this well organized book provides a wealth of data and information vital to anyone involved in environmental exposure assessment groundwater surface water and water contamination pesticide regulation and environmental simulation modeling At your fingertips you will have the latest information on the development of meaningful environmental fate data and how this information will result in accurate assessment of potential environmental and human hazards The inadequacy of current regulatory guidelines and the resulting nonscientific assessment of agrochemical environmental fate are discussed in detail A wide variety of environmental fate studies are included to demonstrate the current use of data to assess environmental fate and potential hazards associated with agrochemical use Finally ten chapters discuss the use of computer models that have been developed for analyzing and integrating data from a variety of environmental fate studies on agrochemicals used under various field conditions

Teaching Secondary Science: A Complete Guide Adam Boxer,2021-11-19 Teaching science is no simple task Science teachers must wrestle with highly abstract and demanding concepts ideas which have taken humanity s greatest minds thousands of years to formulate and refine Communicating these great and awesome theories involves careful forethought and planning We need to deliver crystal clear explanations guide students as they develop their embryonic knowledge and then release them to develop their thinking independently all the while curating and tending to their long term understanding as it develops over time In *Teaching Secondary Science A Complete Guide* Adam breaks down the complex art of teaching science into its component parts providing a concrete and comprehensive set of evidence informed steps to nurturing brilliant science students Adam hopes that you find this book interesting but his main aim is for you to find it useful Useful when it comes to sketching out your curriculum useful when preparing your explanations useful for mapping out how you will check student understanding and useful for all other aspects of science teaching This is a truly complete guide and science teachers of any experience will find it packed with ideas that are new challenging interesting and most importantly useful *Project Earth Science* Alfredo L. Aretxabaleta,Gregg R. Brooks,Nancy W. West,2011 How well can your students

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