

Guided Practice - Acids & Bases**Properties of Acids and Bases**

1. Match the following terms with their descriptions:

- | | | |
|------------------------|-------|--|
| a. Arrhenius Acid | _____ | the substance formed when a proton is added to _____ |
| b. Arrhenius Base | _____ | _____ a base |
| c. Conjugate Acid | _____ | the remaining substance when a proton is lost from an acid |
| d. Conjugate Base | _____ | a substance that produces OH^- in solution |
| e. Bronsted-Lowry Acid | _____ | a substance producing H^+ in solution |
| f. Bronsted-Lowry Base | _____ | a proton acceptor |
| | | a proton donor |

2. Which of the following represent conjugate acid-base pairs? Label each species as a conjugate acid or a conjugate base. For those pairs that are H_2O -T conjugates, write the correct conjugate acid or base for each species in the pair, labeling each species correctly.

- HSO_4^- , SO_4^{2-}
- HBr , BrO^-
- H_3PO_4 , PO_4^{3-}
- H_2O , HO^-

3. Complete the following table regarding acid strength.

Property	Strong Acid	Weak Acid
Definition		
Describe Dissociation		
Species present in solution		
Relative strength of conjugate base compared to water		
Equation showing acid's reaction to water		

Chapter 20 Acids And Bases Answer Key

Wisner



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Workflow
Students will read the pages in their book and then complete each section of the teacher guide. They should be encouraged to complete as many of the activities and projects as possible as well. Tests are given at regular intervals with space to record each grade.

About the Author
DR DENNIS ENGLIN earned his bachelor's from Westmont College, his master of science from California State University, and his EdD from the University of Southern California. He enjoys teaching animal biology, vertebrate biology, wildlife biology, organismic biology, and astronomy at The Master's University. His professional memberships include the Creation Research Society, the American Fisheries Association, Southern California Academy of Sciences, Yellowstone Association, and Au Sable Institute of Environmental Studies.

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Chemistry James Vincent Quagliano, L. M. Vallarino, 1969 *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain), 1912 *Exploring the Physical Sciences* Willard J. Poppy, Leland L. Wilson, 1965

Watts' Dictionary of Chemistry, Revised and Entirely Rewritten Henry Watts, 1905 **Chem 5e Irm** Steven S. Zumdahl, 2000-02 **General Inorganic Chemistry** Mayce Cannon Sneed, 1926 **Chemistry** Steven S. Zumdahl, 1989

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