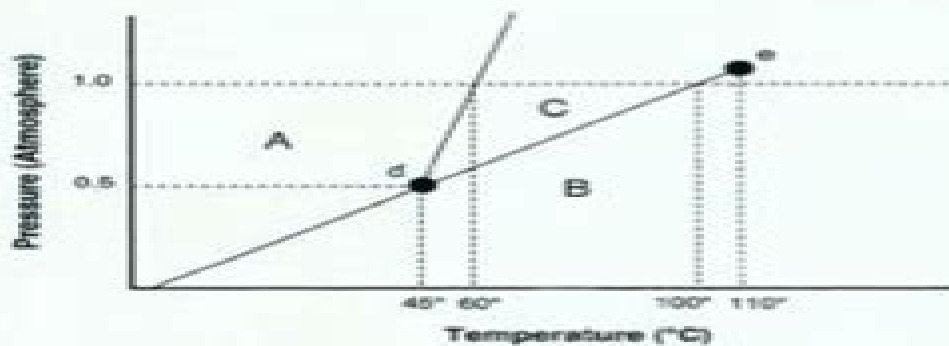


Part A – Generic Phase Diagram.

Answer the questions below in relation to the following generic phase diagram.



1. Which section represents the solid phase? _____
2. What section represents the liquid phase? _____
3. What section represents the gas phase? _____
4. What letter represents the triple point? _____ In your own words, what is the definition of a triple point?
5. What is this substance's normal melting point? _____
6. What is this substance's normal boiling point? _____
7. Above what temperature is it impossible to liquefy this substance, no matter what the pressure? _____
8. At what temperature and pressure do all three phases coexist?

9. At a constant temperature, what would you do to cause this substance to change from the liquid phase to the solid phase?
10. What does sublimation mean?

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RS Peters



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