

DESCRIBING CHEMICAL REACTIONS

Section Review - Objectives

- Explain how to write a word equation
- Describe how to write a skeleton equation
- List the steps for writing a complete chemical equation.

Vocabulary

- chemical equation
- skeleton equation
- catalyst
- coefficients
- balanced equation

Part A Completion

Use this completion exercise to check your understanding of the concepts and terms that are introduced in this section. Each blank can be completed with a term, short phrase, or number.

A chemical reaction can be concisely represented by a chemical 1.

The substances that undergo a chemical change are the 2. The new substances formed in a chemical reaction are the 3. In accordance with the law of conservation of 4, a chemical equation must be balanced.

When balancing an equation, you place 5 in front of reactants and products so that the same number of atoms of each 6 are on each side of the equation. An equation must never be balanced by changing the 7 in the chemical formula of a substance.

Special symbols are used to show the physical state of a substance in a reaction. The symbol for a liquid is 8; for a solid, 9; and for a gas, 10. A substance dissolved in water is designated 11. If a 12 is used to increase the rate of a chemical reaction, its formula is written above the arrow.

1. _____
2. _____
3. _____
4. _____
5. _____
6. _____
7. _____
8. _____
9. _____
10. _____
11. _____
12. _____

Part B True-False

Classify each of these statements as always true, AT; sometimes true, ST; or never true, NT.

- _____ 13. In an equation, a substance is shown to be in the gaseous state by placing an upward-pointing arrow after its formula.
- _____ 14. The symbol Δ placed over the arrow in an equation means that heat is supplied to the reaction.
- _____ 15. Atoms are destroyed in a chemical reaction.
- _____ 16. A skeleton equation is not a balanced equation.

Part D Questions and Problems

24. Write a balanced equation for each of these chemical reactions. I

a. Aluminum reacts with aqueous hydrochloric acid to form hydrogen gas and aqueous aluminum chloride.

b. Acetylene gas (C_2H_2) burns in a welding torch with oxygen to form carbon dioxide gas and water vapor.

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LSENS, a General Chemical Kinetics and Sensitivity Analysis Code for Homogeneous Gas-phase Reactions. 2: Code Description and Usage Krishnan Radhakrishnan,1994 **Chemistry** James N. Spencer,George M. Bodner,Lyman H. Rickard,2010-12-28 CHEMISTRY *Prentice Hall Science Explorer* Michael J. Padilla,2002 *Material Balances for Chemical Reacting Systems* R.L. Cerro,B.G. Higgins,S. Whitaker,2022-12-05 Written for use in the first course of a typical chemical engineering program *Material Balances for Chemical Reacting Systems* introduces and teaches students a rigorous approach to solving the types of macroscopic balance problems they will encounter as chemical engineers This first course is generally taken after students have completed their studies of calculus and vector analysis and these subjects are employed throughout this text Since courses on ordinary differential equations and linear algebra are often taken simultaneously with the first chemical engineering course these subjects are introduced as needed Teaches readers the fundamental concepts associated with macroscopic balance analysis of multicomponent reacting systems Offers a novel and scientifically correct approach to handling chemical reactions Includes an introductory approach to chemical kinetics Features many worked out problems beginning with those that can be solved by hand and ending with those that benefit from the use of computer software This textbook is aimed at undergraduate chemical engineering students but can be used as a reference for graduate students and professional chemical engineers as well as readers from environmental engineering and bioengineering The text features a solutions manual with detailed solutions for all problems as well as PowerPoint lecture slides available to adopting professors Cehmistry Textbook for College and University USA Ibrahim Sikder,2023-06-04 Cehmistry Textbook USA **Physical Chemistry for the Life Sciences** Peter Atkins,Julio de Paula,2011-01-30 Peter Atkins and Julio de Paula offer a fully integrated approach to the study of physical chemistry and biology Text-book of Elementary Chemistry Alexander Smith,1915 **American Journal of Pharmacy and the Sciences Supporting Public Health** ,1895 **Scientific American** ,1904 **The Pharmaceutical Journal and Pharmacist** ,1911 **The Southern California Practitioner** ,1905 An Outline of Physics Albert Edward Caswell,1928 *Report of the Secretary for Public Instruction* ... Queensland. Department of Public Instruction,1912 **Report of the Minister for Education and Cultural Activities** Queensland. Department of Education,1912 Chemical News and Journal of Physical Science ,1879 *Pharmaceutical Journal;* ,1904 Indiana Pharmacist ,1885 **Prentice Hall Chemistry** ,2000 The Chemical News and Journal of Industrial Science ,1919 American Journal of Pharmacy ,1895

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