

Name _____ Date _____ Class _____

CHAPTER 11 REVIEW*Gases***SECTION 2****SHORT ANSWER** Answer the following questions in the space provided.

1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:

_____ a. temperature increases, volume stays the same

_____ b. volume increases, temperature stays the same

_____ c. temperature decreases, volume stays the same

_____ d. volume decreases, temperature stays the same

2. Two sealed flasks, A and B, contain two different gases of equal volume at the same temperature and pressure. Assume that flask A is warmed as flask B is cooled. Will the pressure in the two flasks remain equal? If not, which flask will have the higher pressure?

PROBLEMS Write the answer on the line to the left. Show all your work in the space provided.

3. A bicycle tire is inflated to 55 lb/in.² at 15°C. Assume that the volume of the tire does not change appreciably once it is inflated.

- a. If the tire and the air inside it are heated to 30°C by road friction, does the pressure in the tire increase or decrease? (Assume the volume of air in the tire remains constant.)

- b. Because the temperature has doubled, does the pressure double to 110 psi?

- c. What will the pressure be when the temperature has doubled? Express your answer in pounds per square inch.

Chapter 1review Gases Section Answers

Annelies Wilder-Smith



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