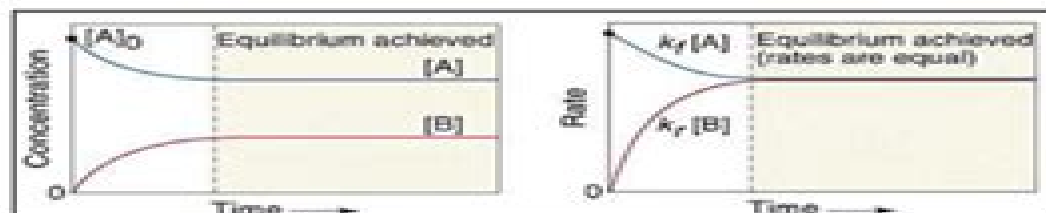


## Chapter 14 Worksheet 1 (ws14.1)

### Chemical Equilibrium, the Equilibrium Constant ( $K_{eq}$ ), and the Reaction Quotient ( $Q$ )

As a reaction proceeds in a **closed system**, the concentrations of reactants and products change until, at **equilibrium**, the rate of the forward reaction is equal to the rate of the reverse reaction. At that point there is **no net change** in the concentrations of reactants or products. For example, for the simple reaction:  $A \rightleftharpoons B$ :



**Regardless of the initial concentrations**, the concentrations of reactants and products present **at equilibrium** are related through the equilibrium constant ( $K_{eq}$ ) as follows:

For the reaction:  $aA + bB \rightleftharpoons cC + dD$

If all reactants and products are in solution then:

$$K_{eq} = K_c = \frac{[C]_{eq}^c [D]_{eq}^d}{[A]_{eq}^a [B]_{eq}^b}$$

(The subscript “eq” indicates “equilibrium” concentration. The subscript “c” stands for “concentration”. The exponents are the coefficients from the balanced equation.)

If all reactants and products are ideal gases then **either** partial pressures or concentrations can be used in the equilibrium constant expression. In terms of partial pressures:

$$K_{eq} = K_p = \frac{(P_C)_{eq}^c (P_D)_{eq}^d}{(P_A)_{eq}^a (P_B)_{eq}^b} \quad \text{(The subscript “p” stands for “pressure”).}$$

The numerical value of the two equilibrium constants above will be different. Remember the ideal gas law shows that the partial pressure of a gas is directly proportional to its molarity:

$$P = \frac{nRT}{V} = MRT \quad (M \text{ is molarity and } R = 0.0821 \text{ L-atm/mol-K}).$$

Using this relationship, it is easy to show that (see pages 605-606 in your textbook) :

$$K_p = K_c(RT)^{\Delta n} \quad \text{where } \Delta n = \text{moles of gaseous products} - \text{moles of gaseous reactants}$$

For reactions involving both gases and aqueous solutions, molar concentrations **and** partial pressures can appear in the equilibrium constant expression.

# Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet

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## **Table of Contents Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet**

1. Understanding the eBook Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - The Rise of Digital Reading Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Personalized Recommendations
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet User Reviews and Ratings
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet and Bestseller Lists
5. Accessing Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet Free and Paid eBooks
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet Public Domain eBooks
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet eBook Subscription Services
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet Budget-Friendly Options
6. Navigating Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet eBook Formats
  - ePub, PDF, MOBI, and More
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet Compatibility with Devices
  - Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Highlighting and Note-Taking Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Interactive Elements Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet

8. Staying Engaged with Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
9. Balancing eBooks and Physical Books Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Setting Reading Goals Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Fact-Checking eBook Content of Chapter 1rotational Equilibrium Exercises Worksheet Answer Sheet
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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