

Introduction to Chemistry

Section 1.1 The Stories of Two Chemicals

In your notebook, read aloud the passage below.

Use each of the terms below just once to complete the passage.

stratosphere
atmosphere

oxygen gas
troposphere

ozone
ultraviolet radiation

ozone hole

Earth's (1) _____ is made up of several layers. The air we breathe makes up the lowest level. This layer is called the (2) _____. The next layer up is called the (3) _____. This level contains a protective (4) _____ layer.

Ozone holes occur when (5) _____ is removed by ultraviolet radiation in the upper part of the atmosphere. The ozone forms a layer around Earth, which absorbs (6) _____. Without ozone, you are more likely to get a sunburn or possibly skin cancer. The thinning of the ozone layer, called the (7) _____, is a problem because without ozone all organisms on Earth are subject to more than one much radiation.

In your notebook, read aloud chlorofluorocarbons.

For each statement below, write true or false.

- _____ (A) CFC is another name for a chlorofluorocarbon.
- _____ (B) CFCs are made up of carbon, fluorine, and oxygen.
- _____ (C) All CFCs are synthetic chemicals.
- _____ (D) CFCs usually react easily with other chemicals.
- _____ (E) CFCs were developed as replacement for toxic refrigerants.

Chapter 1 stoichiometry Study Guide For Content Mastery

Steven S. Zumdahl, Paul B. Kelter



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Resources in Education ,1980 **Research in Education** ,1969 *Study Guide for Chemical Principles [by] Steven S. Zumdahl* Steven S. Zumdahl,Paul B. Kelter,1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Glencoe Chemistry* Glencoe/McGraw-Hill,2002 Student Study Guide/Solutions Manual for Essentials of General, Organic, and Biochemistry Denise Guinn,Rebecca Brewer,Rachel C. Lum,2009-09-15 The Student Study Guide and Solutions Manual provides students with a combined manual designed to help them avoid common mistakes and understand key concepts After a brief review of each section s critical ideas students are taken through stepped out worked examples try it yourself examples and chapter quizzes all structured to reinforce chapter objectives and build problem solving techniques The solutions manual includes detailed solutions to all odd numbered exercises in the text *Chemistry Lesson Plans, Study Guides, and Lecture Notes* Doctor Chemistry,2017-08-13 The Ultimate Guide to Learning or Teaching Chemistry This book contains the real lecture notes and slide of a highly effective high school and college Chemistry teacher Teachers Never plan another lesson again Students Ace your upcoming exam This series covers all of the topics of High School Chemistry and General Chemistry including Accuracy and Significant Figures Mixtures Metric System Bonding Atomic Theory Periodic Table VSEPR Ionic and Covalent Bonding Geometric Bonding The Mole and Molar Mass Equation Balancing Thermodynamics Stoichiometry States of Matter Gas Laws and Calculations Reaction Calculations Acids and Bases Limiting Reagents Redox and Electro Chemistry Organic Chemistry Basics *Principles of Modern Chemistry* Wade A. Freeman,David W. Oxtoby,1996 PRINCIPLES OF MODERN CHEMISTRY has dominated the honors and high mainstream general chemistry courses and is considered the standard for the course The fifth edition is a substantial revision that maintains the rigor of previous editions but reflects the exciting modern developments taking place in chemistry today Authors David W Oxtoby and H P Gillis provide a unique approach to learning chemical principles that emphasizes the total scientific process from observation to application placing general chemistry into a complete perspective for serious minded science and engineering students Chemical principles are illustrated by the use of modern materials comparable to equipment found in the scientific industry Students are therefore exposed to chemistry and its applications beyond the classroom This text is perfect for those instructors who are looking for a more advanced general chemistry textbook

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