

KEY

Use the following to answer questions 1 & 2.



1. If 24 grams of sodium chloride reacts with an excess amount of magnesium oxide, how many grams of sodium oxide will be produced?

A	B
Given: 24 g NaCl = ? grams Na ₂ O	

Conversions: Molar mass of NaCl: 1 mole NaCl = 58 g NaCl
Mole to mole ratio: 2 mole NaCl = 1 mole Na₂O
Molar mass of Na₂O: 1 mole Na₂O = 62 g Na₂O

Setup:

$$24 \text{ g NaCl} \times \frac{1 \text{ mole NaCl}}{58 \text{ grams NaCl}} \times \frac{1 \text{ mole Na}_2\text{O}}{2 \text{ mole NaCl}} \times \frac{62 \text{ g Na}_2\text{O}}{1 \text{ mole Na}_2\text{O}} = 12.8 \text{ g Na}_2\text{O}$$

2. If 53 grams of magnesium oxide reacts with an excess amount of sodium chloride, how many grams of magnesium chloride will be made?

	A	B
Given:	53 g MgO	= ? g MgCl ₂

Conversions: Molar mass of MgO: 1 mole MgO = 40 g MgO
Mole to Mole Ratio : 1 mole MgO = 1 mole MgCl₂
Molar mass of MgCl₂ : 1 mole MgCl₂ = 94 g MgCl₂

Setup:

$$53 \text{ g MgO} \times \frac{1 \text{ mole MgO}}{40 \text{ g MgO}} \times \frac{1 \text{ mole MgCl}_2}{1 \text{ mole MgO}} \times \frac{94 \text{ g MgCl}_2}{1 \text{ mole MgCl}_2} = 124.55 \text{ g MgCl}_2$$

Use the following to answer questions 3 & 4.



3. If 14 moles of potassium oxide reacts with an excess amount of magnesium nitride, how many grams of potassium nitride will be made?

Given: 14 grams $\overset{\text{A}}{\text{K}_2\text{O}} = ? \overset{\text{B}}{\text{g K}_3\text{N}}$

Conversions: Molar mass of K_2O : 1 mole K_2O = 94 g K_2O
Mole to mole ratio: 3 moles K_2O = 2 moles K_3N
Molar mass K_3N : 1 mole K_3N = 131 g K_3N

Setup:

$$14 \text{ grams K}_2\text{O} \times \frac{1 \text{ mole K}_2\text{O}}{94 \text{ g K}_2\text{O}} \times \frac{2 \text{ moles K}_3\text{N}}{3 \text{ moles K}_2\text{O}} \times \frac{131 \text{ g K}_3\text{N}}{1 \text{ mole K}_3\text{N}} = 13 \text{ g K}_3\text{N}$$

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David M. Dunfield



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Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Ebook: Chemistry: The Molecular Nature of Matter and Change** Silberberg, 2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change
Automotive Engine Performance Nicholas Goodnight, Kirk VanGelder, 2019-02-20 Automotive Engine Performance published as part of the CDX Master Automotive Technician Series provides technicians in training with a detailed overview of modern engine technologies and diagnostic strategies Taking a strategy based diagnostic approach it helps students master the skills needed to diagnose and resolve customer concerns correctly on the first attempt Students will gain an understanding of current diagnostic tools and advanced performance systems as they prepare to service the engines of tomorrow **Master Dentistry Volume 3 Oral Biology** Barry Berkovitz, Bernard J. Moxham, Roger W. A. Linden, Alastair J. Sloan, 2010-09-17 Master Dentistry Oral Biology is designed as a revision guide for dental students and offers the curriculum essentials in an easy to digest format Each section is fully illustrated throughout and is supported by extensive self assessment questions which allow the reader to assess their own knowledge of the topic and perfect their exam techniques New to the Master Dentistry series this volume addresses oral anatomy and its related histology physiology and biochemistry and is perfect for both undergraduate students and post graduates preparing for the MJDF Quick reference revision aid for dental students ideal for exam preparation Covers the essentials of the subject to a level that is expected with the GDC s curriculum outlined in the First Five Years document Each chapter provides a brief overview of the topic and lists the essential learning objectives for that area of study Presents key anatomical biochemical and physiological material in a useful integrated clinically relevant format Includes extensive self testing material true false questions extended matching questions picture questions and essay questions enabling readers to assess their knowledge and perfect exam techniques Contains unique mind map summary sheets to provide crucial information in a pictorial format to further promote learning
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Table of Contents Chapter 1study Guide For Content Mastery Stoichiometry Key

1. Understanding the eBook Chapter 1study Guide For Content Mastery Stoichiometry Key
 - The Rise of Digital Reading Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 1study Guide For Content Mastery Stoichiometry Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Personalized Recommendations
 - Chapter 1study Guide For Content Mastery Stoichiometry Key User Reviews and Ratings
 - Chapter 1study Guide For Content Mastery Stoichiometry Key and Bestseller Lists
5. Accessing Chapter 1study Guide For Content Mastery Stoichiometry Key Free and Paid eBooks
 - Chapter 1study Guide For Content Mastery Stoichiometry Key Public Domain eBooks
 - Chapter 1study Guide For Content Mastery Stoichiometry Key eBook Subscription Services

- Chapter 1study Guide For Content Mastery Stoichiometry Key Budget-Friendly Options
- 6. Navigating Chapter 1study Guide For Content Mastery Stoichiometry Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 1study Guide For Content Mastery Stoichiometry Key Compatibility with Devices
 - Chapter 1study Guide For Content Mastery Stoichiometry Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Highlighting and Note-Taking Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Interactive Elements Chapter 1study Guide For Content Mastery Stoichiometry Key
- 8. Staying Engaged with Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 1study Guide For Content Mastery Stoichiometry Key
- 9. Balancing eBooks and Physical Books Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 1study Guide For Content Mastery Stoichiometry Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Setting Reading Goals Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Fact-Checking eBook Content of Chapter 1study Guide For Content Mastery Stoichiometry Key
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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