#### Chapter 20 Review Questions

1R) Which element is reduced in the reaction below?

 $\text{Re}(CO)_5(l) + 2\text{HI}(g) \rightarrow \text{Fe}(CO)_4\text{I}_2(s) + CO(g) + \text{H}_2(g)$ 

- A) For CIO ID9 H
- E) 1.

Approved D

Angiver: B

Answer: C

Acever A

2R) Which element is reduced in the reaction below?

Re2+ + H++ Cr>O>2- → Fe3+ + Cr3+ + H>O

- ATEM: CIO D): H
- 4R) Which of the following reactions is a redox reaction?
  - (a) K<sub>2</sub>CrO<sub>4</sub> + BaCl<sub>2</sub> → BaCrO<sub>4</sub> + 2KCl
  - (b) Pb<sub>2</sub><sup>2+</sup> + 2Br- → PbBr
  - (c) Cu + S → CuS
- A) (a) only B) (b) only
- D) (a) and (c) C) (c) only
- E) (b) and (c)
- 6R) Which substance is the oxidizing agent in the following reaction?

Fe<sub>2</sub>S<sub>3</sub> + 12HNO<sub>3</sub> -- 2Fe<sub>1</sub>NO<sub>3</sub>)<sub>3</sub> + 3S + 6NO<sub>2</sub> + 6H<sub>2</sub>O

- A) HNOs. B) S
- C) NO2 D) Fe<sub>2</sub>S<sub>3</sub>
- E) H<sub>2</sub>O Angwer: A
- 7R) What is the coefficient of the permanganate ion when the following equation is balanced?

 $MnO_4^- + Br^- \rightarrow Mn^{2+} + Br_2$  (acidic solution)

- ANT B) 2
- C) 3 D0.5 E) 4 Angwer: B
- 8R) What is the coefficient of the dichromate ion when the following equation is balanced?

 $Fe^{2+} + Cr_2O_7^{2-} \rightarrow Fe^{3+} + Cr^{3+}$  (acidic solution)

- A)1
- B) 2
- C13: D0.5

E) 6

# **Chapter 20 Review Questions**

Roger LeRoy Miller, Gaylord A. Jentz

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web jul 30 2020 in this context we would like to draw your kind attention towards para 6 of directorate letter no 137 64

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