Nuclear Chemistry

SECTION 1

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SHORT	ANSWER	AUTHORN OF	time:	following	questions	100	TITLE:	space	provided.

- Based on the information about the three elementary particles in the text, which has the greatest mass?
 - (a) the proton
 - (b) the neutron
 - (c) the electron
 - (d) They all have the same mass.
- The force that keeps nucleons together is
 - (a) a strong nuclear force.
 - (b) a weak nuclear force.
 - (c) an electromagnetic force.
 - (d) a gravitational force.
- The stability of a nucleus is most affected by the
 - (a) number of neutrons.
 - (b) number of protons.
 - (c) number of electrons.
 - (d) ratio of neutrons to protons.
- If an atom should form from its constituent particles,
 - (a) matter is lost and energy is taken in.
 - (b) matter is lost and energy is released.
 - (c) matter is gained and energy is taken in.
 - (d) matter is gained and energy is released.
- For atoms of a given mass number, those with greater mass defects, have
 - (a) smaller binding energies per nucleon.
 - (b) greater binding energies per nucleon.
 - (c) the same binding energies per nucleon as those with smaller mass defects.
 - (d) variable binding energies per nucleon.
- Based on Figure 1.1 of the text, which isotope of He, belium-3 or belium-4.
 - a. has the smaller binding energy per nucleon?
 - b. is more stable to nuclear changes?
- 7. The number of neutrons in an atom of magnesium-25 is . . Nuclides of the same element have the same

Chapter 21 Review Nuclear Chemistry Answers

B Lingard

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