

Chapter 25 Nuclear Chemistry Practice Test

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. What does the 4 in ${}^4_2\text{He}$ represent?
- the mass number
 - the atomic number
 - the number of protons
 - the number of neutrons
- ____ 2. What does the 101 in ${}^{256}_{101}\text{Md}$ represent?
- the mass number
 - the atomic number
 - the nuclide number
 - the number of neutrons
- ____ 3. Balance the following equation: ${}^{226}_{88}\text{Rn} \rightarrow {}^{226}_{86}\text{Rn} + \text{_____}$
- ${}^4_2\text{He}$
 - 1_0e
 - ${}^1_1\text{H}$
 - ${}^0_{-1}e$
- ____ 4. Balance the following equation: ${}^{239}_{93}\text{Np} \rightarrow \text{_____} + {}^0_{-1}e$
- ${}^{239}_{90}\text{Th}$
 - ${}^{239}_{92}\text{U}$
 - ${}^{239}_{94}\text{Pu}$
 - ${}^{239}_{94}\text{Pa}$
- ____ 5. The spontaneous disintegration of a nucleus into a slightly lighter and more stable nucleus, accompanied by emission of particles, electromagnetic radiation, or both, is
- nuclear fusion.
 - nuclear radiation.
 - radioactive decay.
 - nuclear fission.
- ____ 6. Which of the following forms of radiation has the greatest penetrating power?
- alpha particles
 - beta particles
 - gamma rays
 - positrons
- ____ 7. Alpha particles are
- electrons.
 - helium nuclei.
 - electromagnetic waves.
 - neutrons.
- ____ 8. Beta particles are
- electrons.
 - helium nuclei.
 - electromagnetic waves.
 - neutrons.
- ____ 9. Which of the illustrations below represents a fission reaction?

Chapter 25 Nuclear Chemistry Answers

Wisner



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Chemistry McGraw-Hill Staff, 2001-03 **Chemistry for the IB Diploma Second Edition** Richard Harwood, Christopher Coates, Christopher Talbot, 2015-07-31 Provide clear guidance to the 2014 changes and ensure in depth study with accessible content directly mapped to the new syllabus and approach to learning This second edition of the highly regarded first edition contains all SL and HL content which is clearly identified throughout Options are available free online along with appendices and data and statistics Improve exam performance with exam style questions including from past papers Integrate Theory of Knowledge into your lessons and provide opportunities for cross curriculum study Stretch more able students with extension activities The shift to concept based approach to learning Nature of Science is covered by providing a framework for the course with points for discussion Key skills and experiments included Full digital package offered in a variety of formats so that you can deliver the course just how you like **Student Study Guide to Accompany Petrucci's General Chemistry** Robert K. Wismer, 1985 *Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook* McGraw Hill, 2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook Matter and Change
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 Chemistry & Chemical Reactivity John C. Kotz,Keith F. Purcell,1987 **Organic Chemistry** Seyhan N. Ege,1998-11

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