

## Nuclear Chemistry Worksheet

Identify the following types of nuclear radiation

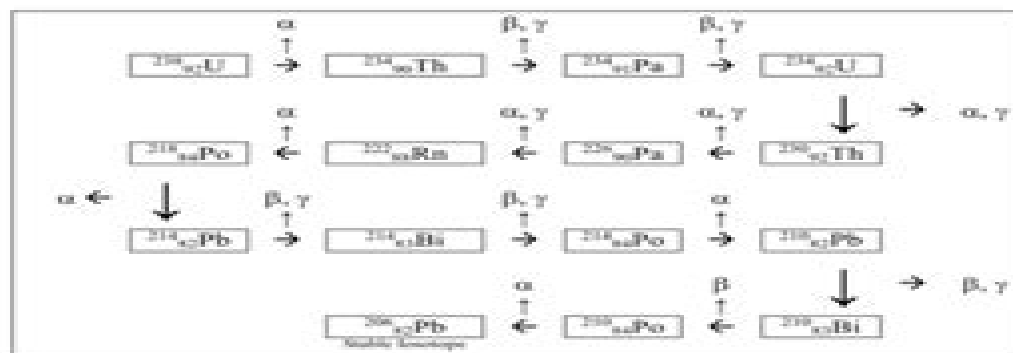
1.  ${}_{-1}^0\text{e}$  \_\_\_\_\_
2.  ${}_{2}^4\text{He}$  \_\_\_\_\_
3.  ${}_{0}^0\gamma$  \_\_\_\_\_
4. Has no mass or charge, almost always follows  $\alpha$  or  $\beta$  decay \_\_\_\_\_
5. Occurs when a neutron breaks apart and forms a proton \_\_\_\_\_
6. Least penetrating nuclear decay \_\_\_\_\_
7. Most damaging nuclear decay to the human body \_\_\_\_\_
8. Nuclear decay that can be stopped by skin or paper. \_\_\_\_\_
9. Nuclear decay that can be stopped by aluminum. \_\_\_\_\_
10. As a result, a proton is transformed into a neutron \_\_\_\_\_
11. The opposite of beta decay \_\_\_\_\_
12. Type of decay isotopes  $>Z83$  will undergo \_\_\_\_\_

Complete the following nuclear equations.



### Nuclear Decay Series

The figure maps the radioactive decay of uranium-238 to lead-206. Use the figure to answer the following questions.



19. How many alpha particles are produced as one atom of uranium-238 decays to an atom of lead-206?  
 \_\_\_\_\_

20. How many beta particles? \_\_\_\_\_

21. Explain why lead-206 is a stable isotope. \_\_\_\_\_

\_\_\_\_\_  
 \_\_\_\_\_  
 \_\_\_\_\_

# Chapter 25 Nuclear Chemistry Worksheet Answer Key

**D Kirk**



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**NUCLEAR**

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