

## Chapter 2: Measurements and Calculations

### 2.1 Scientific Notation

- A quantitative observation is a measurement using numbers
  - Example: a substance weighing 4.3 grams
- Numbers associated with scientific measurements are often large or very small so they are expressed with scientific notation
- Scientific notation expresses a number as a product of a number between 1 and 10 and the appropriate power of 10.

$$93,000,000 = 9.3 \times 10,000,000 = 9.3 \times 10^7$$

Number between 1 and 10      Appropriate power of 10 (10,000,000 =  $10^7$ )

$$93,000,000 = 9.3 \times 10^7$$

We moved the decimal point seven places to the left, so we need  $10^7$  to compensate.

### 2.2 Units

- A Unit is a measurement that tells us the scale/standard that is being used to represent the result of the of measurement
- Two most widely used systems are the English system used in the United States and the metric system used in most of the rest of the industrialized world
- In 1960, the International System was set up which was a comprehensive system of units.

**Table 2.1** Some Fundamental SI Units

Physical Quantity	Name of Unit	Abbreviation
mass	kilogram	kg
length	meter	m
time	second	s
temperature	kelvin	K

- Since fundamental units are not always a convenient size, the International System or SI system uses prefixes to change the size of the unit
- Measurements consist of both a number and a unit

### 2.3 Measurements of Length, Volume, and Mass

- Fundamental SI unit of length is meter (1 meter = 39.37 inches)
- Fractions of a meter or multiples of a meter can be expressed by powers of 10
- Volume is the amount of a 3D space occupied by a substance
- Fundamental unit of volume in the SI is based on a cube that measures 1 meter in each of the three directions

# Chapter 2 Measurements And Calculations Standardized Test Prep Answers

**Tim Collins**



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