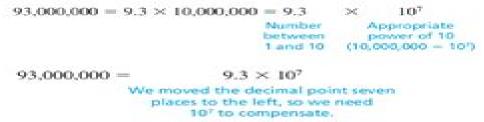
Chapter 2: Measurements and Calculations

- 2.1 Scientific Notation
 - A quantitative observation is a measurement using numbers
 - Example: a substance weighing 4.3 grams
 - Numbers associated with scientific measurements are often large or very small so they are expressed with scientific notation
 - Scientific notation expresses a number as a product of a number between 1 and 10 and the appropriate power of 10.



2.2 Units

- A Unit is a measurement that tells us the scale/standard that is being used to represent the result of the of measurement
- Two most widely used systems are the English system used in the United States and the metric system used in most of the rest of the industrialized world
- In 1960, the International System was set up which was a comprehensive system of units.

	Table	2.1	Some	Fundamental	SI Units
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Physical Quantity	Name of Unit	Abbreviation
mass	kilogram	kg
length	meter	m
time	second	s
temperature	kelvin	K

- Since fundamental units are not always a convenient size, the International System or SI system uses prefixes to change the size of the unit
- Measurements consist of both a number and a unit
- 2.3 Measurements of Length, Volume, and Mass
 - Fundamental SI unit of length is meter (1 meter = 39.37 inches)
 - Fractions of a meter or multiples of a meter can be expressed by powers of 10
 - Volume is the amount of a 3D space occupied by a substance
 - Fundamental unit of volume in the SI is based on a cube that measures 1 meter in each
 of the three directions

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Fitness 5 Body Composition 6 Musculoskeletal Fitness PART III CONDITIONING FOR PHYSICAL FITNESS 7 The Acute and Chronic Effects of Exercise 8 Exercise Prescription 9 Nutrition and Performance PART IV PHYSICAL ACTIVITY AND DISEASE 10 Heart Disease 11 Cancer 12 Diabetes 13 Obesity 14 Psychological Health 15 Aging Osteoporosis and Arthritis

16 Exercise Risks Appendix A Physical Fitness Test Norms Appendix B Calisthenics for Development of Flexibility and Muscular Strength and Endurance Appendix C Major Bones Muscles and Arteries of the Human Body Appendix D Compendium of Physical Activities Glossary Index

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Table of Contents Chapter 2 Measurements And Calculations Standardized Test Prep Answers

- 1. Understanding the eBook Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - The Rise of Digital Reading Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Personalized Recommendations
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers User Reviews and Ratings
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers and Bestseller Lists

- 5. Accessing Chapter 2 Measurements And Calculations Standardized Test Prep Answers Free and Paid eBooks
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers Public Domain eBooks
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers eBook Subscription Services
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers Budget-Friendly Options
- 6. Navigating Chapter 2 Measurements And Calculations Standardized Test Prep Answers eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers Compatibility with Devices
 - Chapter 2 Measurements And Calculations Standardized Test Prep Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Highlighting and Note-Taking Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Interactive Elements Chapter 2 Measurements And Calculations Standardized Test Prep Answers
- 8. Staying Engaged with Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 2 Measurements And Calculations Standardized Test Prep Answers
- 9. Balancing eBooks and Physical Books Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 2 Measurements And Calculations Standardized Test Prep Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Setting Reading Goals Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Fact-Checking eBook Content of Chapter 2 Measurements And Calculations Standardized Test Prep Answers
 - Distinguishing Credible Sources

- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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