

Chapter 1: Chemistry and Measurement

Chemistry (noun): *a science studying the composition, characteristics, properties, and transformations of all matter in the known universe.*

Most modern textbooks start their coverage of chemistry with a discussion of why chemistry is so important to modern life, how it is relevant, why it is the “central science”, blah, blah, blah.

Here is the truth; chemistry is important *to you* because a group of professionals decided that you must study chemistry in order to earn the degree in your major, or to gain admission to a particular kind of professional school. Whether you agree with their decision or not is irrelevant: you have chosen your major, and your major requires you to study chemistry.

If you don't want to study chemistry, then choose a different major. You don't have to be a chemist, (or a biologist, or a physician, or whatever) in order to be happy and successful. You won't hurt my feelings if you choose to be an artist or a business executive. I would prefer you to be a happy and successful business executive, and not know any chemistry at all, than to have you study a subject you detest. Just don't expect me to justify to you why you should study chemistry; I don't owe you a justification, and neither does science in general nor chemistry in particular.

The above definition of chemistry says it all: all matter, anything you can touch or that can touch you, is within the scope of chemistry. Any type or kind of physical substance you can name or describe is subject to the laws of chemistry.

Normally, the next topic would be a lengthy discussion of the scientific method, hypotheses, theories, laws, and so on. I'm not going to indulge in this waste of paper; instead I'll give you two relatively simple definitions and consider this topic finished.

Theory (noun): *a detailed and elaborate explanation for a set of apparently unrelated facts or observations.*

Scientific law (noun): *a principle or pattern observed in nature, and commonly expressed mathematically.*

An example of a scientific law is the *law of conservation of mass*, which tells us that, in a closed system, the total mass is constant. This law is expressed mathematically as:

$$M_{\text{MASS}}(\text{initial_total}) = M_{\text{MASS}}(\text{final_total})$$

Chapter 1chemistry Study Guide

Iris Stovall, Steven S. Zumdahl



Chapter 1 chemistry Study Guide:

Chemistry, Student Study Guide John A. Olmsted, Gregory M. Williams, 2005-02-02 100% Pure Chemical Understanding Every morning many of us are energized by a cup of coffee Imagine if you were as energized by understanding the chemistry in your morning cup from the coffee trees which fill red coffee berries with caffeine and a variety of other chemical substances to the feathery crystals formed by the caffeine molecules to the decaffeinating machines which use liquid solvents to remove this stimulant from some of the beans Now that's real chemical understanding Olmsted and Williams Fourth Edition of Chemistry focuses on helping you see and think about the world and even your coffee as a chemist This text helps you understand how chemical phenomena are governed by what happens at the molecular level apply critical thinking skills to chemical concepts and problems and master the basic mathematical techniques needed for quantitative reasoning You'll see the world as chemists do and learn to appreciate the chemical processes all around us A Fourth Edition with a lot of new perks Revisions include a new early energy chapter revised coverage of bonding expanded coverage of intermolecular forces and increased coverage of multiple equilibria including polyprotic acids New pedagogy strengthens students critical thinking and problem solving skills Visual Summaries at the end of each chapter use molecular and diagrammatic visual elements to summarize essential skills concepts equations and terms eGrade Plus provides an integrated suite of teaching and learning resources including a complete online version of the text links between problems and relevant sections in the online text practice quizzes the Visual Tutor Interactive LearningWare problems and lab demos as well as homework management and presentation features for instructors

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