

ChemThink: Behavior of Gases

Name _____ Per _____

1. What does the speed of atoms depend on? Describe the relationship between speed and each of these factors.

temperature: \uparrow temp = \uparrow speed

type of atom: smaller/lighter = faster

2. In a sample of one type of atom, do all the atoms move at the same speed? Why?

No, speed is related to each individual atoms energy

3. What is the equation for Pressure and what are the units for pressure?

$$\text{Pressure} = \frac{\text{Force}}{\text{Area}} \quad \text{psi}$$

4. When atoms are moving in a closed container what is creating the force on the sides of the container?

impacts (collisions) of atoms w/ sides of container

5. Explain how a change in temperature causes pressure to change and include an equation.

more collisions results in higher pressure

\downarrow
b/c molecules moving faster

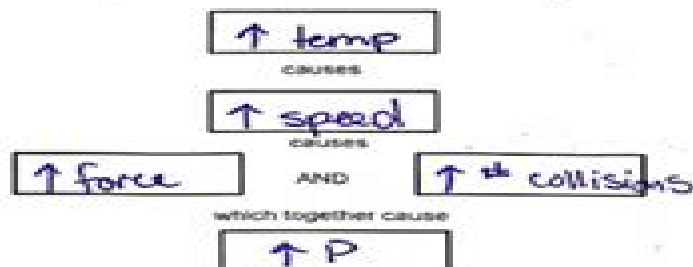
$$\frac{P}{T} = \text{constant}$$

6. Draw the graph you observed of pressure vs. temperature. Is this a direct relationship or an inverse relationship?



direct

7. Fill the cause/effect relationship in the diagram of boxes you observed.



- Explain how the number of atoms present causes pressure to change and include an equation?

more atoms = more collisions, which increases pressure

$$\frac{P}{n} = \text{constant}$$

Chapter 1review Gases Answer Key

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Chapter 1 review Gases Answer Key:

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Table of Contents Chapter 1review Gases Answer Key

1. Understanding the eBook Chapter 1review Gases Answer Key
 - The Rise of Digital Reading Chapter 1review Gases Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chapter 1review Gases Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chapter 1review Gases Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chapter 1review Gases Answer Key
 - Personalized Recommendations
 - Chapter 1review Gases Answer Key User Reviews and Ratings

- Chapter 1review Gases Answer Key and Bestseller Lists
- 5. Accessing Chapter 1review Gases Answer Key Free and Paid eBooks
 - Chapter 1review Gases Answer Key Public Domain eBooks
 - Chapter 1review Gases Answer Key eBook Subscription Services
 - Chapter 1review Gases Answer Key Budget-Friendly Options
- 6. Navigating Chapter 1review Gases Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Chapter 1review Gases Answer Key Compatibility with Devices
 - Chapter 1review Gases Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chapter 1review Gases Answer Key
 - Highlighting and Note-Taking Chapter 1review Gases Answer Key
 - Interactive Elements Chapter 1review Gases Answer Key
- 8. Staying Engaged with Chapter 1review Gases Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chapter 1review Gases Answer Key
- 9. Balancing eBooks and Physical Books Chapter 1review Gases Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chapter 1review Gases Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chapter 1review Gases Answer Key
 - Setting Reading Goals Chapter 1review Gases Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chapter 1review Gases Answer Key
 - Fact-Checking eBook Content of Chapter 1review Gases Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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