#### CHAPTER 11 REVIEW

# Gases

#### SECTION 2

SHORT ANSWER Answer the following questions in the space provided.

- 2. Two sealed flasks, A and B, contain two different gases of equal volume at the same temperature and pressure. Assume that flask A is warmed as flask B is cooled. Will the pressure in the two flasks remain equal? If not, which flask will have the higher pressure?

**PROBLEMS** Write the answer on the line to the left. Show all your work in the space provided.

- A bicycle tire is inflated to 55 lb/in.<sup>2</sup> at 15°C. Assume that the volume of the tire does not change appreciably once it is inflated.
  - a. If the tire and the air inside it are heated to 30°C by road friction, does the pressure in the tire increase or decrease? (Assume the volume of air in the tire remains constant.)
  - b. Because the temperature has doubled, does the pressure double to 110 psi?
  - c. What will the pressure be when the temperature has doubled? Express your answer in pounds per square inch.

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