

Chapter 12

STUDY GUIDE

Mixtures and Solutions

Section 12.1: Homogeneous and Heterogeneous Mixtures

Do what material classification experiments and activities.

The classification table below has an error.

State _____	1. A substance's particles separating from the whole can be observed at the molecular level.
State _____	2. The most abundant substances in nature is the dispersed medium.
State _____	3. A solid is a pure, separately substance.
State _____	4. A solid solution consists of a dispersed phase in a solid.
State _____	5. Mixture is not a homogeneous mixture.
State _____	6. An observed heterogeneous mixture is a mixture.
State _____	7. Because some of the particles of a mixture are dispersed in the dispersed medium, the mixture is a mixture.
State _____	8. A mixture of a solid and a liquid is a mixture.
State _____	9. A mixture of a solid and a liquid is a mixture.
State _____	10. A mixture of a solid and a liquid is a mixture.
State _____	11. A mixture of a solid and a liquid is a mixture.
State _____	12. A mixture of a solid and a liquid is a mixture.
State _____	13. A mixture of a solid and a liquid is a mixture.
State _____	14. A mixture of a solid and a liquid is a mixture.
State _____	15. A mixture of a solid and a liquid is a mixture.
State _____	16. A mixture of a solid and a liquid is a mixture.
State _____	17. A mixture of a solid and a liquid is a mixture.
State _____	18. A mixture of a solid and a liquid is a mixture.
State _____	19. A mixture of a solid and a liquid is a mixture.
State _____	20. A mixture of a solid and a liquid is a mixture.

The table below shows the classification of mixtures as homogeneous and heterogeneous. Place a check in the appropriate classification column for each mixture described.

Mixture Description	Homogeneous	Heterogeneous	Temperature
1. A mixture of a solid and a liquid.			
2. A mixture of a solid and a liquid.			
3. A mixture of a solid and a liquid.			
4. A mixture of a solid and a liquid.			
5. A mixture of a solid and a liquid.			
6. A mixture of a solid and a liquid.			
7. A mixture of a solid and a liquid.			
8. A mixture of a solid and a liquid.			
9. A mixture of a solid and a liquid.			
10. A mixture of a solid and a liquid.			
11. A mixture of a solid and a liquid.			
12. A mixture of a solid and a liquid.			
13. A mixture of a solid and a liquid.			
14. A mixture of a solid and a liquid.			
15. A mixture of a solid and a liquid.			
16. A mixture of a solid and a liquid.			
17. A mixture of a solid and a liquid.			
18. A mixture of a solid and a liquid.			
19. A mixture of a solid and a liquid.			
20. A mixture of a solid and a liquid.			

Chapter 1 study Guide Mixtures And Solutions

**Michael J. Moran, Howard N.
Shapiro, Daisy D. Boettner, Margaret
B. Bailey**

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