

Acids and Bases Worksheet

1. Acids and bases share some properties but are different in others. Use the samples at the side of the class to complete the chart. If both acids and bases share the property, place it in both columns.

Property	Acids	Bases
Conductivity in Aqueous Solution	very	very
Taste	Sour	Bitter
Feel	—	Slippery
Reaction with Metal	produces H_2 gas	—
Reaction with Carbonate	produces CO_2 gas	—
Colour on Litmus Paper	Red	Blue
Colour with Universal Indicator	Red	Purple
Colour with phenolphthalein	Colourless	pink

2. Indicate if each compound is an acid or base and name it.

	Acid/Base?	Name
HNO_3	<u>A</u>	<u>nitric acid</u>
$Mg(OH)_2$	<u>B</u>	<u>magnesium hydroxide</u>
H_3PO_4	<u>A</u>	<u>phosphoric acid</u>
H_2CO_3	<u>A</u>	<u>carbonic acid</u>
HF	<u>A</u>	<u>hydrofluoric acid</u>
$Fe(OH)_3$	<u>B</u>	<u>iron (iii) hydroxide</u>

3. Write the formula for each of the following acids/bases:

hydrochloric acid	<u>HCl</u>	hydrochloric acid	<u>HCl</u>
aluminum hydroxide	<u>$Al(OH)_3$</u>	sodium hydroxide	<u>NaOH</u>
lithium hydroxide	<u>LiOH</u>	calcium hydroxide	<u>$Ca(OH)_2$</u>
sulfuric acid	<u>H_2SO_4</u>	sulfurous acid	<u>H_2SO_3</u>

Chapter 1 test Acids And Bases Answers

RP Manchanda



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Saraswati Chemistry Class 10 RP Manchanda, A text book on Chemistry Analytical Chemistry Bryan M. Ham,Aihui MaHam,2015-10-01 A comprehensive study of analytical chemistry providing the basics of analytical chemistry and introductions to the laboratory Covers the basics of a chemistry lab including lab safety glassware and common instrumentation Covers fundamentals of analytical techniques such as wet chemistry instrumental analyses spectroscopy chromatography FTIR NMR XRF XRD HPLC GC MS Capillary Electrophoresis and proteomics Includes ChemTech an interactive program that contains lesson exercises useful calculators and an interactive periodic table Details Laboratory Information Management System a program used to log in samples input data search samples approve samples and print reports and certificates of analysis *Quick revision to Fundaments of chemistry 10* Amar Bhutani, Quick Revision to Fundamentals of Chemistry Class 10 is a fully exam focused guide crafted to make ICSE Chemistry simple scoring and revision friendly Carefully mapped to the latest syllabus this book provides concept summaries definitions solved numericals balanced equations diagrams and past exam style questions helping students revise quickly and perform with confidence All important chapters Periodic Table Chemical Bonding Acids Bases Salts Metallurgy Organic Chemistry Mole Concept Electrolysis are presented through point wise notes and easy explanations that save time and strengthen retention Features include All textbook questions fully solved Additional ICSE style practice questions Important laws reactions and formulas highlighted Numerical practice with stepwise solutions Revision maps and quick recall charts Perfect for school exams and board revision this guide makes Chemistry predictable manageable and high scoring Arundeepp s Quick Revision series is trusted nationwide for clarity precision and exam success **Chapter-wise NCERT + Exemplar + Practice Questions with Solutions for CBSE Chemistry Class 11 - 2nd Edition** Disha Experts,2017-08-29 The book Chapter wise NCERT Exemplar Practice Questions with Solutions for CBSE Class 11 Chemistry has been divided into 3 parts Part A provides detailed solutions Question by Question of all the questions exercises provided in the NCERT Textbook Part B provides solutions to the questions in the NCERT Exemplar book Part C provides selected Practice Questions useful for the Class 11 examination along with detailed solutions The solutions have been designed in such a manner Step by Step that it would bring 100% Concept Clarity for the student Chemical Primer Simeon Pease Meads,1885 *Questions Set at the Comprehensive Examinations of ...* ,1922 *Questions set at the examinations* College Entrance Examination Board,1922 Study Guide with Selected Solutions Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems **Questions Set at the Examinations** ,1921 *Cooley's Cyclopædia of Practical Receipts and Collateral Information in the Arts, Manufactures, Professions, and Trades, Including Medicine, Pharmacy, and Domestic Economy* Arnold James Cooley,1879 **Cyclopaedia of Practical Receipts**

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