

Name KEYDue date: June 10th 2019

1. List the components of Dalton's Atomic Theory.

- ① All ~~matter~~ ^{matter} is composed of tiny indivisible particles called atoms.
- ② Atoms of any given element share the same chemical properties, while atoms of different elements have different properties.
- ③ Chemical compounds consist of atoms combined in specific ratios.
- ④ Chemical reactions change only the way the atoms are combined in compounds. The atoms themselves are unchanged + do not disappear.

2. Based upon Dalton's Atomic Theory, explain the difference between a sodium atom and a chlorine atom.

A sodium atom + chlorine atom have different chemical properties.

3. Based upon Dalton's Atomic Theory, explain why $Mg_{1.5}Cl$ is not a valid formula.

Atoms must be combined in specific whole number ratios.

4. Complete the following table on the subatomic particles.

PARTICLE	LOCATION	CHARGE
PROTON	nucleus	+ 1
NEUTRON	nucleus	0
ELECTRON	around nucleus	- 1

Chapter 2 Study Guide Answers

Richard O. Straub



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