

Review Unit 12 Test (Chp 20): Electrochemistry

Section I Multiple Choice

1. The loss of electrons by an element is called _____.

A) reduction
B) transduction
C) fractionation
D) oxidation
E) potentiation



2. Which substance oxidizes another substance in the reaction above?

A) Fe_2S_3
B) S
C) NO_2
D) HNO_3
E) H_2O



3. As the reaction represented above proceeds, the oxidation number of manganese changes from...

A) +4 to 0
B) +7 to 0
C) +7 to +4
D) +8 to +4
E) +1 to +2

4. In which of the following species does phosphorus have the same oxidation number as it does in H_2PO_4^- ?

A) P_4O_6
B) P_4O_{10}
C) P_4
D) HPO_3^{2-}
E) H_3PO_3

5. Which one of the following reactions is a redox reaction?

A) $\text{H}_2\text{O} + \text{NaCl} \rightarrow \text{NaOH} + \text{HCl}$
B) $\text{Pb}^{2+} + 2 \text{Cl}^- \rightarrow \text{PbCl}_2$
C) $\text{AgNO}_3 + \text{HCl} \rightarrow \text{HNO}_3 + \text{AgCl}$
D) $\text{NaOH} + \text{HCl} \rightarrow \text{NaCl} + \text{H}_2\text{O}$

Chapter 20 Electrochemistry Test

James P. Birk, Donald M. Baird



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Fundamentals of Electrochemistry V. S. Bagotzky, 1993 A rigorous outline of the basic concepts phenomena processes laws forming the subject matter of modern theoretical and applied electrochemistry originally published in Russian in 1988 by Khimiya Press Moscow In the present English edition three supplementary chapters have been added on photo

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