

# Chapter 2: Acids, Bases and Salts

Science Worksheet

	Standard: 10th	Subject: Science	
Student Name:			

Q1. Match the chemical substances given in Column (A) with their appropriate application given in Column (B):

	Column (A)		Column (8)	
Α.	Bleaching powder	1.	Preparation of glass	
В.	Baking soda	2.	Production of H <sub>2</sub> and Cl <sub>2</sub>	
C.	Washing soda	3.	Decolourisation	
D.	Sodium chloride	4.	Antacid	

- 1. A-2, B-1, C-4, D-3
- 2. A-3, B-2, C-4, D-1
- 3. A-3, B-4, C-1, D-2
- 4. A-2, B-4, C-1, D-3
- Q2. 10ml, of a solution of NaOH is found to be completely neutralized by 8ml, of a given solution of HCI. If we take 20ml, of same solution of NaOH, the amount of HCI solution required to neutralize it will be:
  - 1. 4mL
  - 2. 8mL
  - 3. 12mL
  - 4. 16mt.
- Q3. Calcium phosphate is present in tooth enamel. Its nature is:
  - 1. Basic.
  - 2. Acidic.
  - 3. Neutral.
  - 4. Amphoteric.
- Q4. Which of the following salts will give an aqueous solution having pH of almost 7?
  - 1. NH<sub>a</sub>NO<sub>3</sub>
  - 2. NH<sub>2</sub>CI
  - 3. CaCl<sub>2</sub>.



# **Chapter 1review Acids And Bases Section Answers**

**Tobias Bleicker** 

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