Acids, Bases, and Salts Review

			ny compound dissociating s. Example: HNO,(aq.) →	
	solution to form ions. Example: $HNO_3(aq.) \rightarrow H^*(aq.) + NO_3(aq.)$. Complete the following equation:			
	HCI (aq.) →			
@	An Arrhenius base is defined as any compound dissociating in an aqueous solution to form ions. Example: KOH (aq) → K [*] (aq) + OH [*] (aq). Complete the following equation: NaOH (aq.) →			
3	Salts are compounds that dissociate in an aqueous solution releasing neither nor ions. Example: KCl (aq.) → K* (aq.) + Cl* (aq.).			
	Complete the following equation:			
		NaCl (aq.) →	<u> </u>	
④	A solution that has strong conductivity and reacts with the metal to make hydrogen gas is a			
	Strong Base	b Weak Base	Strong Acid	d Weak Acid
	Answers			
3	Which of the following is slippery to touch?			
	Strong Acid	b Strong Base	Strong Electrolyte	d Water
	Answers			
•	The hydronium ion is			
	■ H ₃ O.	B H₂O	C OH-	d H ₂
	Answers			
0	A substance that changes color in an acidic or basic solution is called			
	Neutralizer	b Inhibitor	pH indicator	d Catalyst
	Answers			
(8)	Name the following acids and bases			
	HF		HNO3	
	кон		Mg(OH),	

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Raymond E. Davis, Kenneth W. Whitten

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