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Chapter Test B

Chapter: Chemical Bonding

| PARTI | On the | line at the | left of eac | h estatuement, | welling the | Better of | the r | cholice | Photo: |
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- 1. An ionic bond results from electrical attraction between
 - is, cuttons and antons.
 - les, automores,
 - 4. dipoles.
 - di antistasi.
- A nonpolar covalent bond is artificily when two atoms of different elements join because the atoms are likely to differ in
 - at demailing
 - b, setains of resultion.
 - c. electronogativity:
 - d. polarity.
- 3. Bond longth is the distance between two bunded atoms at
 - Oheir minimum potential energy.
 - b, their reactinum kinetic energy,
 - c. their maximum potential energy.
 - d. one-half the diameter of the electron cloud.
- _____ 4. To draw a Lewis structure, it is not necessary to know
 - which atoms are in the replecule.
 - b, bond energies.
 - c, the number of valence electrons for each store.
 - 4. the number of atoms is the molecule.
- S. For multiple covalent bonds to form in molecules, the molecules must contain carbon, nitrogen, or
 - as eddications.
 - b. Bydrogen.
 - e. oncygon.
 - d. beitrere.
- 6. The principle that states that atoms tend to form compounds in which each atom has eight electrons in its highest occupied energy level is called the
 - as make of elightsu-
 - by configuration rule.
 - c. Avogadro principle.
 - d. occord rube.

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