

**CHEM 1411 – General Chemistry I**  
**Practice Problems, Chapters 1–3**

**Chapter 1 – Chemistry: The Study of Change**

- 1.** Element, compound, homogeneous mixture (solution), or heterogeneous mixture:
- |                                       |                       |   |
|---------------------------------------|-----------------------|---|
| a) orange juice<br>(freshly-squeezed) | b) brass              | c) 0.9% saline (NaCl) solution                |
| d) garden soil                        | e) room air           | f) methane gas                                |
| g) sodium metal                       | h) N <sub>2</sub> gas | i) Cu(NO <sub>3</sub> ) <sub>2</sub> crystals |
- 2.** Define (some of these terms are found in Chapters 2 and 3):
- |                         |                                |
|-------------------------|--------------------------------|
| a) element              | b) compound                    |
| c) solution             | d) physical change             |
| e) chemical change      | f) law of conservation of mass |
| g) intensive properties | h) extensive properties        |
| i) precision            | j) accuracy                    |
| k) isotopes             | l) stoichiometry               |
| m) the mole             | n) Avogadro's number           |
| o) limiting reactant    |                                |
- 3.** Convert
- |  |
|--|
| a) 5 feet 10 inches to meters                |
| b) 55 miles per hour to cm per second        |
| c) 5.0 quarts to liters                      |
| d) 1.00 ft <sup>2</sup> to cm <sup>2</sup>   |
| e) 32.06 amu to g                            |
| f) 25 g of carbon to moles of carbon         |
| g) 1.00 g of gold to number of atoms of gold |
| h) 25 °F to Kelvins                          |
- 4.** How many significant figures are in the following numbers?
- |             |  |               |
|-------------|--|---------------|
| a) 3.1416   | b) 6.022 × 10 <sup>23</sup> atoms/mole | c) 0.000104 g |
| d) 6.0 g/mL | e) 1.00794 amu                         | f) 5000 mi    |

# Chapter 1supplemental Problems Chemistry Answers

**Carolina Biological Supply Company**



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lessons or teaching yourself CliffsStudySolver guides can help you make the grade Author Steven Bassett started teaching anatomy and physiology at the high school level in 1978 He has been the lead instructor for anatomy and physiology at Southeast Community College in Lincoln Nebraska since 1990 He is adjunct professor in the Physician s Assistance Program at Union College in Lincoln

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