

Chapter 20 Electricity

WordWise

Match each definition with the correct term in the grid and then write its number under the appropriate term. When you have filled in all the boxes, add up the numbers in each column, row, and the two diagonals.

What is surprising about the sums? _____

Definitions

1. A property that causes subatomic particles such as protons and electrons to attract or repel other matter
2. The attraction or repulsion between electrically charged objects
3. Charge transfer without contact between materials
4. Law that total charge in an isolated system is constant
5. A continuous flow of electric charge
6. Material through which charge can easily flow
7. Material through which a charge cannot easily flow
8. The opposition to the flow of charges in a material
9. A circuit in which the charge has only one path through which it can flow
10. An electric circuit with two or more paths through which charge can flow
11. A switch that opens when current in a circuit is too high
12. Information sent as patterns in the controlled flow of electrons through a circuit
13. A smoothly varying signal produced by continuously changing the voltage or current in a circuit
14. A complete path through which a charge can flow
15. A solid-state component with three layers of semiconductors
16. A thin slice of silicon that contains many solid-state components

				Diagonal = _____
integrated circuit _____	induction _____	electric force _____	analog signal _____	= _____
electric current _____	parallel circuit _____	circuit breaker _____	resistance _____	= _____
series circuit _____	electrical conductor _____	electrical insulator _____	electronic signal _____	= _____
law of conservation of charge _____	transistor _____	electric circuit _____	electric charge _____	= _____
= _____	= _____	= _____	= _____	Diagonal = _____

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Brendan G. Carr



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web figure 3 1 1 3 1 1 matter is usually classified into three classical states with plasma sometimes added as a fourth state from left to right quartz solid water liquid nitrogen dioxide gas the state a given substance exhibits is also a physical property

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web all matter has physical and chemical properties physical properties are characteristics that scientists can measure without changing the composition of the sample under study such as mass color

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web feb 11 2016 such knowledge of the domain structure of an actual material is extremely important in controlling and tailoring the magnetic properties of the material this chapter is devoted to describing the formation of the domain structures of wires in tandem with their magnetisation processes and their magnetic characteristics

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web structure and properties of matter it all matters and it s all matter chemistry is the branch of science that focuses on the composition of matter the atomic basics the first subatomic particle to be discovered was the electron electrons have a negative electric charge the periodic table

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web properties of matter fall into one of two categories if the property depends on the amount of matter present it is an extensive property the mass and volume of a substance are examples of extensive properties for instance a gallon of milk has

a larger mass and volume than a cup of milk