Energy

Radiation

Particle:

Chapter 25 - Nuclear Chemistry

The study of _____ and their uses in chemistry

Radiation -

The penetrating rays & particles emitted by a radioactive source

Radioactivity -

Process that occurs when nuclei change

Radioisotopes -

- Atoms containing
 - Recall... What are isotopes?

100

How do nuclear reactions differ from chemical reactions?

Chemical Reactions

- Atoms become stable by ______or
 - electrons
- Rate _____changed by catalysts, temperature, pressure, etc.
- Mass & charge ______

Nuclear Reactions

 Nuclei become stable by undergoing changes that ______of energy

Radioactive Atom

- Rate _______be changed
- Charge conserved, mass
 - A ____small amount of mass is converted into energy

Nuclear Stability

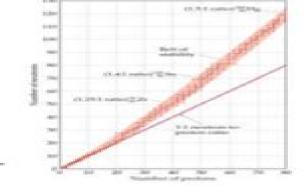
- Most nuclei are stable
- · The stability of a nucleus depends on its neutron to proton
- The stable nuclei are in the '

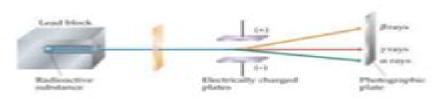
Radioactive Decay

- Occurs in atoms with too ______ or too _____ neutrons
- The nucleus will attempt to become more stable by releasing energy through
- Transmutation -
 - when the unstable nucleus of one element is _______
 into a stable nucleus of a different element
 - Occurs naturally and artificially

Types of Radioactive Decay

- Alpha particles,
- o Beta particles,
- Positron emission
- Gamma rays,





<u>Chapter 25 Nuclear Chemistry Guided Reading And Study Workbook</u>

Léo Pomar

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