

- (a) Define Half-life period. Establish the relation between $t_{1/2}$ and k of the first order reaction.

ଅର୍ଦ୍ଧ ଅବସ୍ଥାବେଳା ବାହା କେଉଁ । ଅର୍ଦ୍ଧ ଅବସ୍ଥାବେଳା ($t_{1/2}$) ଓ k ର ସମ୍ପର୍କ (x) ନିମ୍ନରେ ସ୍ପଷ୍ଟ କରିବା ।

- (f) Distinguish between Molarity and Normality, ଲେଭୋଲିଟି ଓ ନର୍ମାଲିଟି ମଧ୍ୟରେ ପାରସ୍ୟ ପର୍ଯ୍ୟାପ୍ତ ।

- (g) An organic compound 'A' gives hydrogen on reaction with sodium metal. It also gives iodoform test and forms an aldehyde 'B' of molecular formula C_2H_4O on oxidation. Name the compounds A and B.

କେହିଏ ଯେକ ଯେକିକ 'A' ସୋଡ଼ିୟମ ଧାତୁ ସହିତ ପ୍ରତିକ୍ରିୟା କରି ହାଇଡ୍ରୋଜେନ୍ ଉତ୍ପାଦିତ କରେ । ଏହା ମଧ୍ୟ ଆୟୋଡ଼ୋଫର୍ମ ପରୀକ୍ଷା ଦିଏ ଏବଂ ଏହା ଉତ୍ତେଜିତ ହୋଇ ଏକ ଆଲଡିହାଇଡ୍ 'B' ଉତ୍ପାଦିତ କରେ ଯାହାର ଆଣବିକ ସୂତ୍ର C_2H_4O ଅଟେ । ଯେକିକ A ଓ B ନାମ ଦିଅ ।

- (h) The complex $Co(NH_3)_5(CO_3)Cl$ has two ionization isomers. Write their structures and IUPAC names.

$Co(NH_3)_5(CO_3)Cl$ ର ଦୁଇଟି ଆୟୋନୀକ ଇସୋମର ଅଛି । ସେମାନଙ୍କର ସରଳ ଓ IUPAC ନାମ ଲେଖ ।

- (i) Write two important functions of proteins. ପ୍ରୋଟିନ୍‌ର ଦୁଇଟି ଗୁରୁତ୍ୱପୂର୍ଣ୍ଣ ଫଳନ ଲେଖ ।
(j) Give one chemical test to distinguish between Acetaldehyde and Acetone.

ଏସିଟାଲଡିହାଇଡ୍ ଏବଂ ଏସିଟୋନ୍ ମଧ୍ୟରେ ପ୍ରତ୍ୟେକ ରାସାୟନିକ ପରୀକ୍ଷା ଦିଅ ।

4. Answer any seven questions of the following :

3×7 = 21

ନିମ୍ନଲିଖିତ ମଧ୍ୟରୁ ଯେକୌଣସି ସାତଟି ପ୍ରଶ୍ନର ଉତ୍ତର ଦିଅ :

- (a) Derive the Nernst equation of electrode potential at $25^\circ C$ for the electrode reaction



$M^{n+}_{(aq)} + ne \rightleftharpoons M_{(s)}$ ଏହି ଚଳିତ ପ୍ରତିକ୍ରିୟାର $25^\circ C$ ରେ ଚଳିତ ଶକ୍ତିର ନର୍ନଷ୍ଟ ସମୀକରଣ ଉତ୍ପାଦନ କର ।



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