

3

Biochemistry and Molecular Analysis

LEARNING OBJECTIVES

After reading this chapter, you will be able to understand the following:

- Method of analysis of chemical composition of a tissue.
- Carbohydrates and their types.
- Amino acids and their classification.
- Proteins and their functions.
- Nucleic acids and their components.
- Lipids and their types.
- Types of bonds linking monomers in a polymer.
- Qualitative tests of biomolecules.

All living and non-living things consist of matter, which can be anything that occupies space and has mass. Mass is the amount of matter in any object, which does not change while weight, the force of gravity acting on matter, does change. Matter exists in three states as solid, liquid and gas. All forms of matter—both living and non-living—are made up of a limited number of building blocks called chemical elements.

3.1 Chemical Composition of Living Forms

Elemental analysis reveals the presence of several elements in both living and non-living matter. Twenty-six elements are present in the human body; four elements, called the major elements constitute about 96% of the body's mass—Oxygen, carbon, hydrogen and nitrogen. Eight others, the lesser elements contribute 3.6% to the body's mass—Calcium (Ca), phosphorus (P), potassium (K), sulphur (S), sodium, chlorine (Cl), magnesium (Mg), and iron (Fe). An additional 14 elements called the trace elements are present in tiny amounts (0.4%)—Aluminum (Al), boron (B), chromium (Cr), cobalt (Co), copper (Cu), fluorine (F), iodine (I), manganese (Mn), molybdenum (Mo), selenium (Se), silicon (Si), tin (Sn), vanadium (V), and zinc (Zn).

All living organisms are made up of same chemicals though their relative amounts may be different. Most of the chemicals in your body exist in the form of compounds. Biologists and chemists divide these compounds into two principal classes:

1. **Inorganic compounds:** They usually lack carbon and are structurally simple. Their molecules also have only a few atoms and cannot be used by cells to perform complicated biological functions. They include Na, K, Ca, Mg, water and many salts of (NaCl, CaCO_3), acids and bases, compounds of sulphate and phosphate. They may have either ionic or covalent bonds. Water makes up 55–60% of a lean adult's total body mass.
2. **Organic compounds:** They always contain carbon and usually hydrogen, and always have covalent bonds. Most of them are large molecules and many are made up of long chains of carbon atoms. Organic compounds make up the remaining 38–43% of the human body. Many organic molecules

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