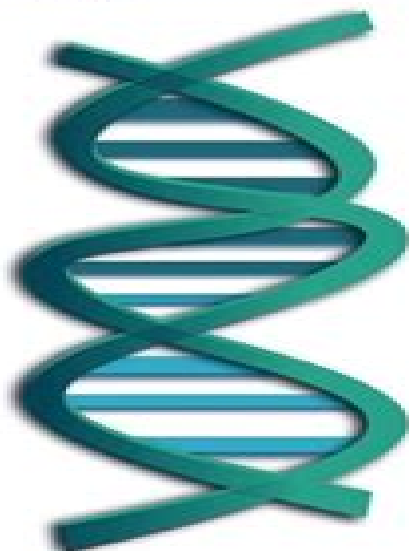


# BIOCHEMISTRY

## PRINCIPLES



### Biochemistry

The word "biochemistry" was first proposed in 1903 by Carl Nerg, a German chemist. Biochemistry is the study of chemical processes in living organisms. Biochemistry governs all living organisms and living processes. Today the main focus of pure biochemistry is in understanding how biological molecules give rise to the processes that occur within living cells which in turn relates greatly to the study and understanding of whole organisms.

### Biomolecules

The four main classes of molecules in biochemistry are carbohydrates, lipids, proteins, and nucleic acids. Many biological molecules are polymers: in this terminology, monomers are relatively small micromolecules that are linked together to create large macromolecules, which are known as polymers. When monomers are linked together to synthesize a biological polymer, they undergo a process called dehydration synthesis.

### AMINO ACID

Amino acids are natural monomers that polymerize at ribosomes to form proteins. Nucleotides, monomers found in the cell nucleus, polymerize to form nucleic acids – DNA and RNA. Glucose monomers can polymerize to form starches, glycogen or cellulose; xylose monomers can polymerize to form xylan. In all these cases, a hydrogen atom and a hydroxyl (-OH) group are lost to form H<sub>2</sub>O, and an oxygen atom links each monomer unit. Due to the formation of water as one of the products, these reactions are known as dehydration or condensation reactions.

### Conventions and nomenclature

#### Nucleic acids

The convention for a nucleic acid sequence is to list the nucleotides as they occur from the 5' end to the 3' end of the polymer chain, where 5' and 3' refer to the numbering of carbons around the ribose ring which participate in forming the phosphate diester linkages of the chain. Such a sequence is called the primary structure of the biopolymer.

#### Sugars

Sugar polymers can be linear or branched and are typically joined with glycosidic bonds. However, the exact placement of the linkage can vary and the orientation of the linking functional groups is also important, resulting in  $\alpha$ - and  $\beta$ -glycosidic bonds with numbering definitive of the linking carbons' location in the ring.

#### Lipids

Lipids are chiefly fatty acid esters, and are the basic building blocks of biological membranes. Another biological role is energy storage (e.g., triglycerides). Most lipids consist of a polar or hydrophilic head (typically glycerol) and one to three nonpolar or hydrophobic fatty acid tails, and therefore they are amphiphilic. Fatty acids consist of unbranched chains of carbon atoms that are connected by single bonds alone (saturated fatty acids) or by both single and double bonds (unsaturated fatty acids). The chains are usually 14-24 carbon groups long, but it is always an even number. For lipids present in biological membranes, the hydrophilic head is from one of three classes: Glycolipids, whose heads contain an oligosaccharide with 1-15 saccharide residues. Phospholipids, whose heads contain a positively charged group that is linked to the tail by a negatively charged phosphate group. Sterols, whose heads contain a planar steroid ring, for example, cholesterol. Other lipids include prostaglandins and leukotrienes which are both 20-carbon fatty acyl units synthesized from arachidonic acid. They are also known as fatty acids.

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## **Table of Contents Biochemistry Principles Speedy Study Guides Speedy Publishing**

1. Understanding the eBook Biochemistry Principles Speedy Study Guides Speedy Publishing
  - The Rise of Digital Reading Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Advantages of eBooks Over Traditional Books
2. Identifying Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Biochemistry Principles Speedy Study Guides Speedy Publishing
  - User-Friendly Interface
4. Exploring eBook Recommendations from Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Personalized Recommendations
  - Biochemistry Principles Speedy Study Guides Speedy Publishing User Reviews and Ratings
  - Biochemistry Principles Speedy Study Guides Speedy Publishing and Bestseller Lists
5. Accessing Biochemistry Principles Speedy Study Guides Speedy Publishing Free and Paid eBooks
  - Biochemistry Principles Speedy Study Guides Speedy Publishing Public Domain eBooks
  - Biochemistry Principles Speedy Study Guides Speedy Publishing eBook Subscription Services

- Biochemistry Principles Speedy Study Guides Speedy Publishing Budget-Friendly Options
- 6. Navigating Biochemistry Principles Speedy Study Guides Speedy Publishing eBook Formats
  - ePub, PDF, MOBI, and More
  - Biochemistry Principles Speedy Study Guides Speedy Publishing Compatibility with Devices
  - Biochemistry Principles Speedy Study Guides Speedy Publishing Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Highlighting and Note-Taking Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Interactive Elements Biochemistry Principles Speedy Study Guides Speedy Publishing
- 8. Staying Engaged with Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Biochemistry Principles Speedy Study Guides Speedy Publishing
- 9. Balancing eBooks and Physical Books Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Biochemistry Principles Speedy Study Guides Speedy Publishing
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Setting Reading Goals Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Fact-Checking eBook Content of Biochemistry Principles Speedy Study Guides Speedy Publishing
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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