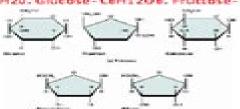
The exam will be primarily multiple-choice, as well as possibly multiple answer and matching. The exam covers all material from the PowerPoint lectures over biological molecules, cells and cell structure, prokaryotes, and eukaryotes and their evolution (material from textbook Chapters 3, 4 and 15). You are responsible for the material covered in the PowerPoint lectures and textbook. Below is a list of topics, but I may inadvertently leave something out. IF IT IS DISCUSSED IN YOUR NOTES AND TEXTBOOK, YOUR ARE RESPONSIBLE FOR THAT INFORMATION ON THE EXAM. Your lecture notes are your true study guide.

Chemistry of Life - Chapter 3 - Biomolecules

- 1. Organic molecules
 - a. What are the 4 most common elements in living organisms? carbon, bydrogen, oxygen, and nitrogen.
 - What defines an organic molecule (in the chemistry sense)? Be able to recognize if a molecule is organic or inorganic (not organic), molecules that are carbon based and hydrogen
 - c. Carbon is the key element in organic molecules
 - How many electrons are in carbon's valence shell?
 - II. How many covalent bonds can carbon form? 4
 - III. What is the other element found in all organic molecules? hydrogen
 - d. Organic Biomolecules: What is a biomolecule? Chemical compounds / molecules produced by a living organism. What are the four major classes of biomolecules? Carbs, proteins, nucleic acid, lipids
- 2. Carbohydrates: Know the ratio of carbon, hydrogen and oxygen atoms in carbohydrates. 1 carbon; 2 hydrogen; 1 oxygen For example, if a sugar has 6 carbons, how many hydrogen atoms and oxygen atoms would it have? 12 hydrogen 6 oxygen Know the different types and examples of carbohydrates. Know the functions of the different types of carbohydrates. Monosaccharides—store and produce energy. EX: glucose, disaccharides—provide energy and to help with the absorption of nutrients. Broken down into two monosaccharides during digestion and they are then used for energy or to help build other molecules like proteins and lipids, polysaccharides provide structural support, storing energy, and sending cellular communication signals. EX: starch, glycogen, cellulose and chitin. Store energy with cells within a body. Be able to recognize monosaccharides, disaccharides and polysaccharides in an illustration.
 - a. Monosactharides ("simple sugars" made of one sugar):
 - Examples: 6-carbon monosaccharides include glucose, fructose, & galactose what is the chemical formula for these monosaccharides? General formula- CH20. Glucose - C6H1206. Fructose - C6H1206



- II. Be able to recognize a monosaccharide in an illustration.
- b. Disaccharides (simple sugars made of "two sugars" meaning 2 monosaccharides bonded together)
 - Examples: sucrose (cane sugar or table sugar), lactose (milk sugar).

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Marcel A. Müller

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