

Equations Biochemistry

Section	Equation Name	Equation	Use / Description
Amino Acids, Peptides, and Proteins	Isoelectric point (pI)	$\frac{pK_{a, \text{amino group}} + pK_{a, \text{carboxyl group}}}{2}$ $pI_{\text{neutral aa}} = i$ $pI_{\text{acidic aa}} = \frac{pK_{a, \text{side group}} + pK_{a, \text{carboxyl group}}}{2}$ $\frac{pK_{a, \text{amino group}} + pK_{a, \text{side group}}}{2}$ $pI_{\text{basic aa}} = i$	pI = the pH at which the molecule is electrically neutral Aa's with non-ionizable side chains have pI around 6 Aa's with acidic side chains have relatively low isoelectric points (well below 6) Aa's with basic side chains have relatively high isoelectric points (well above 6)
		$E + S \xrightleftharpoons[k_{-1}]{k_1} ES \xrightarrow{k_{cat}} E + P$	k_1 = rate E-S complexes forms k_{-1} = rate E-S complexes dissociate k_{cat} = rate E-S complex turns into E + P
		$v = \frac{v_{max} [S]}{K_m + [S]}$	v = velocity of enzyme v_{max} = maximum enzyme velocity K_m = Michaelis constant (the [S] at which 1/2 E's active sites are full)
		$v_{max} = [E] k_{cat}$	v_{max} = maximum enzyme velocity k_{cat} = # of S molecules "turned over" (converted to product)
	Catalytic efficiency	$\frac{k_{cat}}{K_m}$	k_{cat} = # of S molecules "turned over" (converted to product) K_m = Michaelis constant (the [S] at which 1/2 E's active sites are full)

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