

**Concept-Development
Practice Page**

32-1

Coulomb's Law

1. The diagram is of a hydrogen atom.

- Label the proton in the nucleus with a + sign and the orbital electron with a - sign.
- The electrical interaction between the nucleus and the orbital electron is a force of _____ (attraction) _____ (repulsion).
- According to Coulomb's law,

$$F = k \frac{q_1 q_2}{r^2}$$

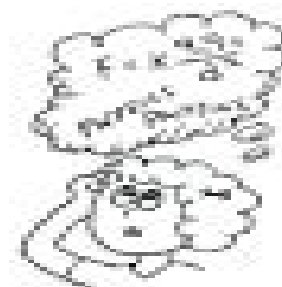
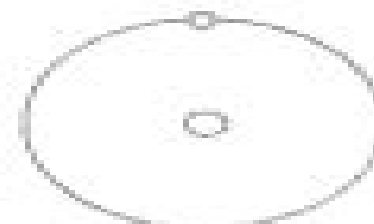
if the charge of either the nucleus or the orbital electron were greater, the force between the nucleus and the electron would be _____ (greater) _____ (less).

and if the distance between the nucleus and electron were greater the force would be _____ (greater) _____ (less).

If the distance between the nucleus and electron were doubled, the force would be _____ (1/4 as much) _____ (1/2 as much) _____ (two times as much) _____ (four times as much).

2. Consider the electric force between a pair of charged particles a certain distance apart. By Coulomb's law:

- If the charge on one of the particles is doubled, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If instead the charge on both particles is doubled, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If instead the distance between the particles is halved, the force is _____ (unchanged) _____ (halved) _____ (doubled) _____ (quadrupled).
- If the distance is halved and the charge of both particles is doubled, the force is _____ times as great.



Concept Development Practice Page 32 Answer Key

WJ Hussar



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Working with Gaumont cameramen and cameras and the new technical advances for the projection of film she became one of the film pioneers ushering in the new era of motion pictures as a narrative form Written by cinema history experts and curators this handsome volume brings to light a critical new mass of Guy Blach s film oeuvre in an effort to restore her to her rightful place in film history **Automotive Engineering** ,1920 **Journal of the Society of Automotive Engineers**

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