



Question:

Heat refers to

- a. the average temperature of a substance.
- b. the total energy of the atoms and molecules in a substance.
- c. the kinetic energy of the atoms and molecules in a substance.
- d. the energy transferred from or to a substance
- e. a value measured by a thermometer.

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Answer:

The correct answer is *d*. Heat refers to energy transferred from or to an object or substance, by any number of means.

“Heat” is often confused with “temperature”—they are two separate, but related, quantities. Heat added *to* a body (measured in Joules, or perhaps calories), typically causes the average kinetic energy of the molecules in that body to increase, thus raising the temperature. Heat removed from a body causes a decrease in the average kinetic energy of the molecules, and thus lowers the temperature. The temperature of the substance may be measured with a thermometer, but this measure—usually indicated in Kelvins or degrees Celsius—is not directly a measure of energy.

In the graphic above, energy is added to the kettle and the water in it by the flame underneath. This causes the internal energy of the kettle and water to increase. Evidence of the increased energy of the water can be seen in the thermometer, which measures the increased temperature (not heat!) of the water. The relationship between heat added and increased temperature is summarized in the relationship $Q = mc\Delta T$.

Conceptual Physics Temperature

University of Northern Iowa

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the University of Northern Iowa text.

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Is There a Temperature? Tamás Sándor Biró, 2011-02-03 Temperature and heat entropy and order or disorder are key classical concepts of physics These are challenged by searching matter under extreme conditions such as high relativistic energy strong acceleration or gravitation or unusual complexity due to long range correlations In our quest for quark matter all these conditions might occur simultaneously This book strongly motivated by the authors everyday research experiences in the field of high energy heavy ion collisions aims to bundle these challenges to modern physics The main topic is at the heart of thermodynamics the very concept of temperature its use and extensions New developments on this issue are both applications and foundations of non extensive statistics as well as concepts borrowed from gravity and string theory to describe the surprisingly statistical behavior of elementary matter at the highest accelerator energies of the world The reader will benefit from bringing these new developments in one book together by having the view of classical and modern concepts at the heart of physics across the problems related to high energy high acceleration and high complexity After reviewing the classical approaches the author discusses the dual gravity and non extensive statistical aspects of heavy ion collisions describing these experimental findings with the use of the concept of temperature

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