

Chemistry Study Guide

Periodic Table of the Elements

| | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
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| <div>1 H</div> | <div>2 He</div> | | | | | | | | | | | | | | | | | <div>18 Ar</div> | <div>19 K</div> | <div>20 Ca</div> | | | | | | | | | | | <div>36 Kr</div> | <div>37 Rb</div> | <div>38 Sr</div> | | | | | | | | | | | <div>54 Xe</div> | <div>55 Cs</div> | <div>56 Ba</div> | | | | | | | | | | | | | | | <div>86 Rn</div> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| <div>3 Li</div> | <div>4 Be</div> | | | | | | | | | | | | | | | | | <div>12 Mg</div> | <div>13 Al</div> | <div>14 Si</div> | <div>15 P</div> | <div>16 S</div> | <div>17 Cl</div> | <div>18 Ar</div> | <div>19 K</div> | <div>20 Ca</div> | <div>21 Sc</div> | <div>22 Ti</div> | <div>23 V</div> | <div>24 Cr</div> | <div>25 Mn</div> | <div>26 Fe</div> | <div>27 Co</div> | <div>28 Ni</div> | <div>29 Cu</div> | <div>30 Zn</div> | <div>31 Ga</div> | <div>32 Ge</div> | <div>33 As</div> | <div>34 Se</div> | <div>35 Br</div> | <div>36 Kr</div> | <div>37 Rb</div> | <div>38 Sr</div> | <div>39 Y</div> | <div>40 Zr</div> | <div>41 Nb</div> | <div>42 Mo</div> | <div>43 Tc</div> | <div>44 Ru</div> | <div>45 Rh</div> | <div>46 Pd</div> | <div>47 Ag</div> | <div>48 Cd</div> | <div>49 In</div> | <div>50 Sn</div> | <div>51 Sb</div> | <div>52 Te</div> | <div>53 I</div> | <div>54 Xe</div> | <div>55 Cs</div> | <div>56 Ba</div> | <div>57 La</div> | <div>58 Ce</div> | <div>59 Pr</div> | <div>60 Nd</div> | <div>61 Pm</div> | <div>62 Sm</div> | <div>63 Eu</div> | <div>64 Gd</div> | <div>65 Tb</div> | <div>66 Dy</div> | <div>67 Ho</div> | <div>68 Er</div> | <div>69 Tm</div> | <div>70 Yb</div> | <div>71 Lu</div> | <div>72 Hf</div> | <div>73 Ta</div> | <div>74 W</div> | <div>75 Re</div> | <div>76 Os</div> | <div>77 Ir</div> | <div>78 Pt</div> | <div>79 Au</div> | <div>80 Hg</div> | <div>81 Tl</div> | <div>82 Pb</div> | <div>83 Bi</div> | <div>84 Po</div> | <div>85 At</div> | <div>86 Rn</div> |
| <div>57 La</div> | <div>58 Ce</div> | <div>59 Pr</div> | <div>60 Nd</div> | <div>61 Pm</div> | <div>62 Sm</div> | <div>63 Eu</div> | <div>64 Gd</div> | <div>65 Tb</div> | <div>66 Dy</div> | <div>67 Ho</div> | <div>68 Er</div> | <div>69 Tm</div> | <div>70 Yb</div> | <div>71 Lu</div> | <div>72 Hf</div> | <div>73 Ta</div> | <div>74 W</div> | <div>75 Re</div> | <div>76 Os</div> | <div>77 Ir</div> | <div>78 Pt</div> | <div>79 Au</div> | <div>80 Hg</div> | <div>81 Tl</div> | <div>82 Pb</div> | <div>83 Bi</div> | <div>84 Po</div> | <div>85 At</div> | <div>86 Rn</div> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |
| <div>87 Fr</div> | <div>88 Ra</div> | <div>89 Ac</div> | <div>90 Th</div> | <div>91 Pa</div> | <div>92 U</div> | <div>93 Np</div> | <div>94 Pu</div> | <div>95 Am</div> | <div>96 Cm</div> | <div>97 Bk</div> | <div>98 Cf</div> | <div>99 Es</div> | <div>100 Fm</div> | <div>101 Md</div> | <div>102 No</div> | <div>103 Lr</div> | <div>104 Rf</div> | <div>105 Db</div> | <div>106 Sg</div> | <div>107 Bh</div> | <div>108 Hs</div> | <div>109 Mt</div> | <div>110 Ds</div> | <div>111 Rg</div> | <div>112 Cn</div> | <div>113 Nh</div> | <div>114 Fl</div> | <div>115 Mc</div> | <div>116 Lv</div> | <div>117 Ts</div> | <div>118 Og</div> | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | | |

🔍 Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles 🧪

💡 Matter:

Anything that has **mass** and **occupies space** 🧪🌐

Examples: air, gases, liquids, solids!

🔬 Atoms:

Atoms are the **smallest, indivisible particle** 🧪👉

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

📌 Subatomic Particle Table:

| Particle | Symbol | Mass (amu) | Charge | Location |
|----------|--------|------------|--------|-------------|
| Proton | p^+ | 1 | +1 | Nucleus |
| Neutron | n | 1 | 0 | Nucleus |
| Electron | e^- | ~ 0 | -1 | Orbit/Shell |

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James R. Braun, Sherman



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