

## Concentration and Molarity PhET Labs

Name: \_\_\_\_\_

**Procedure:** [https://phet.colorado.edu/sims/html/concentration/latest/concentration\\_en.html](https://phet.colorado.edu/sims/html/concentration/latest/concentration_en.html)

### Part 1: Dissolution and Saturation

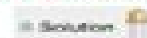
Take some time to play and familiarize yourself with the simulation. Click on everything. Move all the sliders. Notice what happens to the concentration as solid solute is added and when evaporation occurs.



Concentration

Concentration  
(mol/L)  
0.179

- How does the concentration change as solid solute is added? concentration increases
  - How does the concentration change as additional water is added? concentration decreases
  - How does the concentration change as evaporation occurs? concentration increases
  - How do you know when a solution is saturated? concentration does not change (it also shows on screen)
  - When a solution is saturated, and additional solid solute is added, what happens? solid does not dissolve and settles at bottom
  - Why do you think this is? solvent can't hold any more solute
  - How does adding this additional solute change the concentration of this saturated solution? decreases concentration
  - How does evaporation change the concentration of a saturated solution? no change- solution is already saturated
- Part 2: Concentrated Solutions (same website as part 1 just click on the Solution option)**
10. Adding a concentrated solution ... describe a way to determine the concentration of the solution in the spigot. Write your plan here: empty the container and add only one solution from the dropper and then use the sensor to measure concentration



- Using your plan ... how might you get that concentrated solution to become saturated? you can evaporate
- Does your plan work for all the other solutions too? yes Why? / Why Not? evaporation reduces the amount of water (or solute) so concentration increases



### Part 3: Molarity [https://phet.colorado.edu/sims/html/molarity/latest/molarity\\_en.html](https://phet.colorado.edu/sims/html/molarity/latest/molarity_en.html)

Molarity (M) =  $\frac{\text{moles of solute (mol)}}{\text{volume of solution (L)}}$

Molarity is moles per liter, that is, how many moles of solute (entire salt) is dissolved per liter of solution.

First, determine the saturation concentration of each of the solutions, that is, how concentrated can you get each solution before the solution is saturated. If you can't determine the concentration using the simulation "Molarity", try using the simulation "Concentration" (You will use this information again in Part 5, if your instructor requires it)



Molarity

Cobalt (II) nitrate	5.640 M	Potassium chromate	3.350 M
Cobalt chloride	4.330 M	Nickel (II) chloride	5.210 M
Potassium dichromate	0.510 M	Copper sulfate	1.380 M
Gold (III) chloride	2.30 M	Potassium permanganate	0.480 M

# Concentration And Molarity Post Lab Exercises Answers

**VM Jensen**



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**Laboratory Manual for Biotechnology** Verma, Ashish S./ Das Surajit & Singh Anchal, Laboratory Manual in Biotechnology Students Methods in Biotechnology Seung-Beom Hong, M. Bazlur Rashid, Lory Z. Santiago-Vázquez, 2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods Methods in Biotechnology is an invaluable resource for those students and professionals Methods in Biotechnology engages the reader by implementing an active learning approach provided advanced study questions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level Methods in Biotechnology Advanced Methods in Biotechnology I and Advanced Methods in Biotechnology II Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field

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## **Table of Contents Concentration And Molarity Post Lab Exercises Answers**

1. Understanding the eBook Concentration And Molarity Post Lab Exercises Answers
  - The Rise of Digital Reading Concentration And Molarity Post Lab Exercises Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Concentration And Molarity Post Lab Exercises Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Concentration And Molarity Post Lab Exercises Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Concentration And Molarity Post Lab Exercises Answers
  - Personalized Recommendations
  - Concentration And Molarity Post Lab Exercises Answers User Reviews and Ratings
  - Concentration And Molarity Post Lab Exercises Answers and Bestseller Lists
5. Accessing Concentration And Molarity Post Lab Exercises Answers Free and Paid eBooks
  - Concentration And Molarity Post Lab Exercises Answers Public Domain eBooks
  - Concentration And Molarity Post Lab Exercises Answers eBook Subscription Services
  - Concentration And Molarity Post Lab Exercises Answers Budget-Friendly Options
6. Navigating Concentration And Molarity Post Lab Exercises Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Concentration And Molarity Post Lab Exercises Answers Compatibility with Devices
  - Concentration And Molarity Post Lab Exercises Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Concentration And Molarity Post Lab Exercises Answers
  - Highlighting and Note-Taking Concentration And Molarity Post Lab Exercises Answers
  - Interactive Elements Concentration And Molarity Post Lab Exercises Answers

8. Staying Engaged with Concentration And Molarity Post Lab Exercises Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Concentration And Molarity Post Lab Exercises Answers
9. Balancing eBooks and Physical Books Concentration And Molarity Post Lab Exercises Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Concentration And Molarity Post Lab Exercises Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Concentration And Molarity Post Lab Exercises Answers
  - Setting Reading Goals Concentration And Molarity Post Lab Exercises Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Concentration And Molarity Post Lab Exercises Answers
  - Fact-Checking eBook Content of Concentration And Molarity Post Lab Exercises Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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