

SOLUTIONS REVIEW SHEET KEY

① a) $M = \frac{\text{mol}}{\text{L}} = \frac{\left(36.9 \text{ g Na}_2\text{SO}_4 \times \frac{1 \text{ mol}}{142.04 \text{ g}} \right)}{.250 \text{ L}} = 1.04 \text{ M}$

b) $m = \frac{\text{mol}}{\text{kg H}_2\text{O}} = \frac{(.25978 \dots \text{ mol})}{.2356 \text{ kg H}_2\text{O}} = 1.10 \text{ m}$

mass soln = $250 \text{ mL} \times \frac{1.09 \text{ g}}{\text{mL}} = 272.5 \text{ g} - 36.9 \text{ g} = 235.6 \text{ g H}_2\text{O}$
↑
mass solute

c) $\% = \frac{\text{mass Na}_2\text{SO}_4}{\text{total mass}} \times 100 = \frac{36.9 \text{ g}}{272.5 \text{ g}} \times 100 = 13.5 \%$

d) $X_{\text{Na}_2\text{SO}_4} = \frac{(.25978 \dots \text{ mol})}{(.25978 \dots + 13.077 \dots)} = .0195$
↑
 $235.6 \text{ g H}_2\text{O} \times \frac{1 \text{ mol H}_2\text{O}}{18.016 \text{ g}} = 13.077 \dots$



$[\text{SO}_4^{2-}] = 1.04 \text{ M}$

$[\text{Na}^+] = 2.08 \text{ M}$

② Weigh out 2.5 mol of NaCl \Rightarrow 146.1 g. Add this to a 1-L volumetric flask, then dilute to a total volume of 1 L with distilled water.

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LM Reder



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Review Questions and Answers for Veterinary Technicians E-Book Heather Prendergast, 2021-02-01 Prepare for VTNE success *Review Questions and Answers for Veterinary Technicians 6th Edition* provides 5 000 VTNE style questions that have been reviewed and updated to reflect the latest changes to the Veterinary Technician National Examination The book begins with multiple choice questions on basic knowledge including anatomy and physiology hospital management calculations and terminology It continues with a Q A review of core subjects such as pharmacology surgical nursing laboratory procedures diagnostic imaging and pain management Written by veterinary technology educator Heather Prendergast this review includes an Evolve website allowing you to create customized timed practice exams that mirror the VTNE experience More than 5 000 multiple choice questions are rigorously reviewed mirror the type of questions found on the VTNE and are designed to test factual knowledge reasoning skills and clinical judgment Detailed rationales are included in the print text and on the Evolve website reinforcing student knowledge and providing the reasoning behind answers Organization of the book into primary subject areas reflects the latest version of the VTNE Customized exam generator on Evolve offers a simulated test taking experience with customized practice tests and timed practice exams with instant feedback and extended rationales NEW More than 200 new questions are added to this edition Study Guide and Solutions Manual for Seager/Slabaugh's Chemistry for Today Seager/Slabaugh, 2004 The fifth edition of the Study Guide and Student Solutions Manual has been updated to reflect all of the changes to the text This ancillary tests the student on the learning objectives in each chapter and provides answers to all of the even numbered end of chapter exercises New additional activities have been added to include a review of each section of the chapter and a section entitled Tying It All Together with a Laboratory Application **Prentice Hall Chemistry** ,2000 A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 *A Problem Solving Approach to Aquatic Chemistry* Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of *A Problem Solving Approach to Aquatic Chemistry* provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter's main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed in each subsequent chapter illustrating the principles of that chapter In addition each chapter has an Historical Note exploring connections with the people and cultures connected to topics in the text *A Problem*

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