# Math Skills, continued

- 2. Write down the specific heat equation. energy =  $cm\Delta T$
- 3. Substitute the specific heat, mass, and temperature change values, and solve.

energy = 
$$\left(\frac{4,186 \text{ J}}{\text{kg} \cdot \text{K}}\right) \times (200 \text{ kg}) \times (12 \text{ K})$$

energy = 
$$10,000,000 \text{ J} = 1.0 \times 10^4 \text{ kJ}$$







# **Concept Review Using Heat Math Skills Continued**

Xiaolong Qi

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