Concept Review

Section: Compound Names and Formulas

1. Explain the difference between iron(II) nitrate and iron(III) nitrate. What is the significance of the Roman numerals? 2. Name the following ionic compounds, keeping in mind that a transition metal cation must include its charge. a. TiO2 b. BaCl₂ _____ c. CuCl₃ d. KI e. SrCl₂ f. CuBr₂ 3. Describe how covalent compounds are named. 4. Write the chemical formulas for the following compounds: a. lithium oxide (ionic) b, carbon monoxide (covalent) c. carbon tetrachloride (covalent) d. nitrogen trifluoride (covalent) e. calcium chloride (ionic) Contrast molecular formulas and empirical formulas.

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Laurence Booth, W. Sean Cleary, Ian Rakita

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Creighton, Frank Thilly, Gustavus Watts Cunningham, 1922 An international journal of general philosophy Student Solutions Manual for Swokowski/Cole's Algebra and Trigonometry with Analytic Geometry (Classic Edition), 11th Swokowski/Cole, Earl Swokowski, Jeffery A. Cole, 2005-08 The student solutions manual provides worked out solutions to the odd numbered problems in the text Engineering Mechanics, Dynamics Carleton G. Fanger, 1970 The American Ecclesiastical Review Herman Joseph Heuser, 1940 Mathematical Reviews, 2007

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