: <u> </u>



"Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class.

 What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_2H_2

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: H-H

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: H:H

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the timer of bonds is: a single bond is tun shared electrons, a

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Y Pai

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